



## Article

# New arsenate minerals from the Arsenatnaya fumarole, Tolbachik volcano, Kamchatka, Russia. XX. Evseevite, $\text{Na}_2\text{Mg}(\text{AsO}_4)\text{F}$ , the first natural arsenate with an antiperovskite structure

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### Abstract

The new mineral evseevite was found in the Arsenatnaya fumarole, Second scoria cone of the Northern Breakthrough of the Great Tolbachik Fissure Eruption, Tolbachik volcano, Kamchatka, Russia. Evseevite is represented by two chemical varieties. The variety close to the end-member  $\text{Na}_2\text{Mg}(\text{AsO}_4)\text{F}$  (holotype) is associated with sanidine, hematite, tenorite, aegirine, cassiterite, sylvite, halite, johillerite, badalovite, calciojohillerite, hatertite, arsmirandite, yurmarinite, axelite, polyarsite, apththitalite, potassic-magnesium-fluorarfvedsonite, litidionite, ferrisanidine and tridymite. The P- and S-enriched variety (cotype) is associated with hematite, fluorophlogopite, svabite, fluorapatite, tilasite, calciojohillerite, forsterite, cassiterite, belomarinaite and apththitalite. Evseevite occurs as prismatic, acicular or hair-like crystals up to 0.7 mm long combined in clusters up to 0.5 mm, brushes or crusts up to  $2 \times 2$  mm. It is transparent, colourless or pale pinkish, with vitreous lustre.  $D_{\text{calc}}$  is  $3.377 \text{ g cm}^{-3}$  for the holotype and  $3.226 \text{ g cm}^{-3}$  for the cotype. Evseevite is optically uniaxial (+),  $\alpha = 1.545(2)$ ,  $\beta = 1.546(2)$ ,  $\gamma = 1.549(2)$  and  $2V_{\text{meas}} = 40(10)^\circ$ . The empirical formulae calculated based on O+F = 5 apfu are  $(\text{Na}_{1.99}\text{Ca}_{0.03}\text{K}_{0.01})_{\Sigma 2.03}(\text{Mg}_{0.98}\text{Fe}_{0.01}^{3+}\text{Zn}_{0.01}\text{Cu}_{0.01})_{\Sigma 1.01}[(\text{As}_{0.98}\text{Si}_{0.01})_{\Sigma 1.01}\text{O}_4](\text{F}_{0.97}\text{O}_{0.03})$  for the holotype and  $\text{Na}_{2.02}(\text{Mg}_{1.00}\text{Fe}_{0.03}^{3+})_{\Sigma 1.03}[(\text{As}_{0.69}\text{P}_{0.25}\text{S}_{0.07})_{\Sigma 1.01}\text{O}_4](\text{F}_{0.78}\text{O}_{0.22})$  for the cotype. Evseevite is orthorhombic, *Pbcn*,  $a = 5.3224(1)$ ,  $b = 14.1255(3)$ ,  $c = 12.0047(3)$  Å,  $V = 902.53(4)$  Å<sup>3</sup> and  $Z = 8$ . Strong reflections of the powder XRD pattern [ $d, \text{Å}(I)(hkl)$ ] are: 4.001(100)(121), 3.479(56)(023), 3.041(45)(042), 2.657(44)(200), 2.642(68)(142) and 2.613(36)(104). The crystal structure was solved from single-crystal XRD data and refined on powder data by the Rietveld method,  $R_{\text{wp}} = 0.0068$ ,  $R_{\text{p}} = 0.0047$  and  $R_{\text{obs}} = 0.0435$ . Evseevite is isostructural to moraskoite  $\text{Na}_2\text{Mg}(\text{PO}_4)\text{F}$ . The structure of evseevite can be described in terms of anion-centred polyhedra. F-centred octahedra  $[\text{FN}_4\text{Mg}_2]^{7+}$  share faces to form chains  $[\text{FN}_2\text{Mg}]^{3+}$  and  $\text{AsO}_4$  tetrahedra are located between the chains. Evseevite belongs to a small set of minerals with antiperovskite structures and is the first natural arsenate with antiperovskite units. The mineral is named in honour of the Russian mineralogist Aleksandr Andreevich Evseev (born 1949).

**Keywords:** evseevite; new mineral; sodium magnesium fluorarsenate; crystal structure; anion-centred octahedron; antiperovskite structure; fumarole sublimate; Tolbachik volcano

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### Introduction

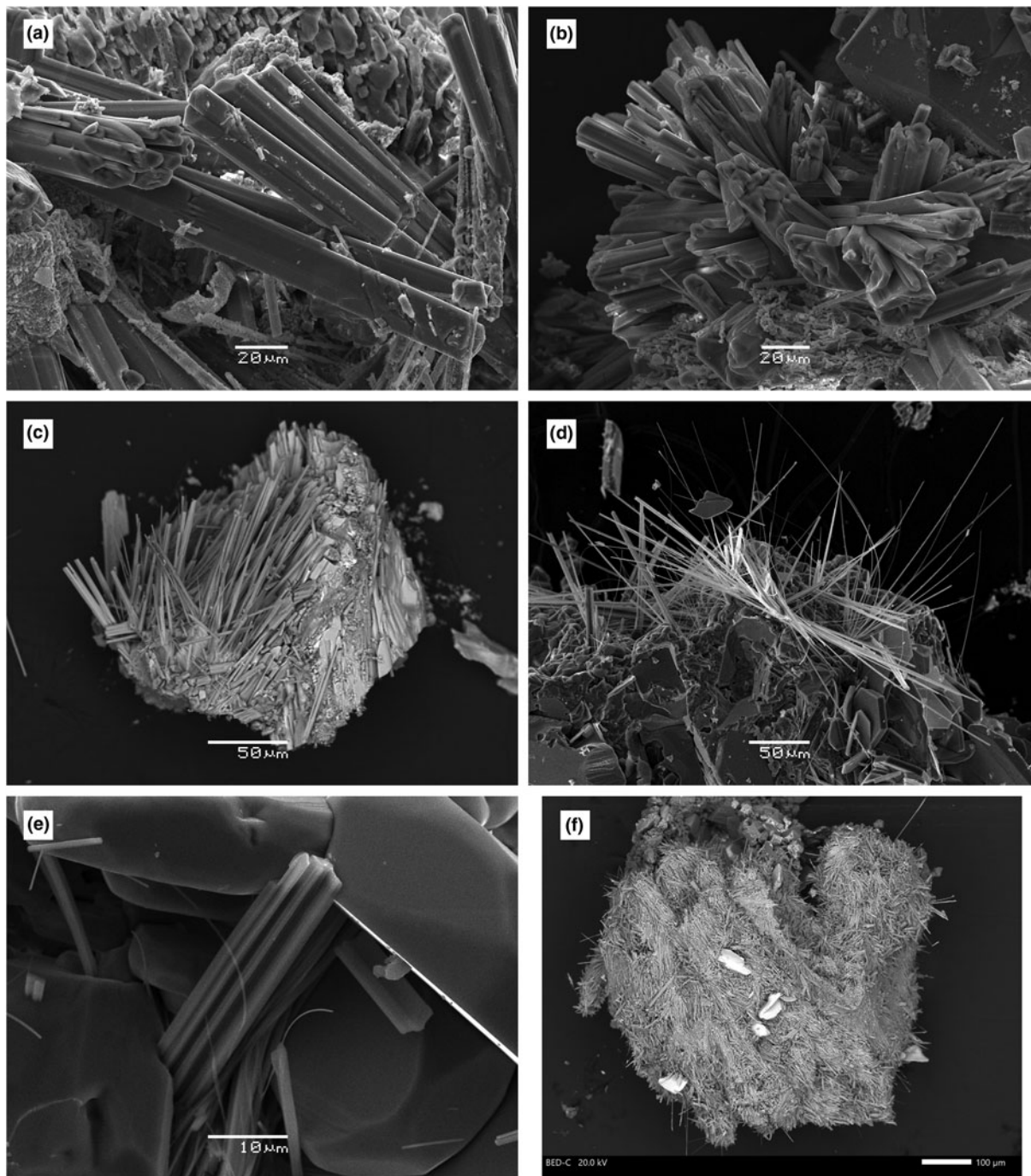
This paper continues the series of articles in which we characterise new arsenates from the Arsenatnaya fumarole at the Second scoria cone of the Northern Breakthrough of the Great Tolbachik Fissure Eruption 1975–1976, Tolbachik volcano, Kamchatka Peninsula, Far-Eastern Region, Russia. Twenty two new mineral species have been described in the previous papers of the series: yurmarinite  $\text{Na}_7(\text{Fe}^{3+}, \text{Mg}, \text{Cu})_4(\text{AsO}_4)_6$  (Pekov *et al.*, 2014a), two polymorphs of  $\text{Cu}_4\text{O}(\text{AsO}_4)_2$ , ericlxmanite and kozyrevskeite (Pekov

*et al.*, 2014b), popovite  $\text{Cu}_5\text{O}_2(\text{AsO}_4)_2$  (Pekov *et al.*, 2015b), structurally related shchurovskyite  $\text{K}_2\text{CaCu}_6\text{O}_2(\text{AsO}_4)_4$  and dmsikolovite  $\text{K}_3\text{Cu}_5\text{AlO}_2(\text{AsO}_4)_4$  (Pekov *et al.*, 2015a), katiarsite  $\text{KTiO}(\text{AsO}_4)$  (Pekov *et al.*, 2016b), melanarsite  $\text{K}_3\text{Cu}_7\text{Fe}^{3+}\text{O}_4(\text{AsO}_4)_4$  (Pekov *et al.*, 2016a), pharmazincite  $\text{KZnAsO}_4$  (Pekov *et al.*, 2017), arsenowagnerite  $\text{Mg}_2(\text{AsO}_4)\text{F}$  (Pekov *et al.*, 2018b), arsenatrotitanite  $\text{NaTiO}(\text{AsO}_4)$  (Pekov *et al.*, 2019d), the two isostructural minerals edtolite  $\text{K}_2\text{NaCu}_5\text{Fe}^{3+}\text{O}_2(\text{AsO}_4)_4$  and alumoedtolite  $\text{K}_2\text{NaCu}_5\text{AlO}_2(\text{AsO}_4)_4$  (Pekov *et al.*, 2019e), anatolyite  $\text{Na}_6(\text{Ca}, \text{Na})(\text{Mg}, \text{Fe}^{3+})_3\text{Al}(\text{AsO}_4)_6$  (Pekov *et al.*, 2019b), zubkovaite  $\text{Ca}_3\text{Cu}_3(\text{AsO}_4)_4$  (Pekov *et al.*, 2019a), pansnerite  $\text{K}_3\text{Na}_3\text{Fe}_6^{3+}(\text{AsO}_4)_8$  (Pekov *et al.*, 2020c), badalovite  $\text{NaNaMg}(\text{MgFe}^{3+})(\text{AsO}_4)_3$  (Pekov *et al.*, 2020b), calciojohillerite  $\text{NaCaMgMg}_2(\text{AsO}_4)_3$  (Pekov *et al.*, 2021a), yurgensonite  $\text{K}_2\text{SnTiO}_2(\text{AsO}_4)_2$  (Pekov *et al.*, 2021b), paraberzeliite  $\text{NaCaCaMg}_2(\text{AsO}_4)_3$  (Pekov *et al.*, 2022a), khrenovite  $\text{Na}_3\text{Fe}_2^{3+}(\text{AsO}_4)_3$  (Pekov *et al.*, 2022b), and axelite  $\text{Na}_{14}\text{Cu}_7(\text{AsO}_4)_8\text{F}_2\text{Cl}_2$  (Pekov *et al.*, 2023).

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**Figure 1.** Morphology of evseevite: (a–c: holotype, #6328) – aggregates of prismatic to acicular crystals overgrowing badalovite (b – with large hematite crystal); (d–e: cotype, #6260) – long-prismatic to acicular and hair-like crystals on hematite and fluorophlogopite; f – sample #7529, pilous crust completely covering calciojohillerite crystals (with bright white tenorite crystals). SEM images, SE (a, b, d, e) and BSE (c, f) modes.

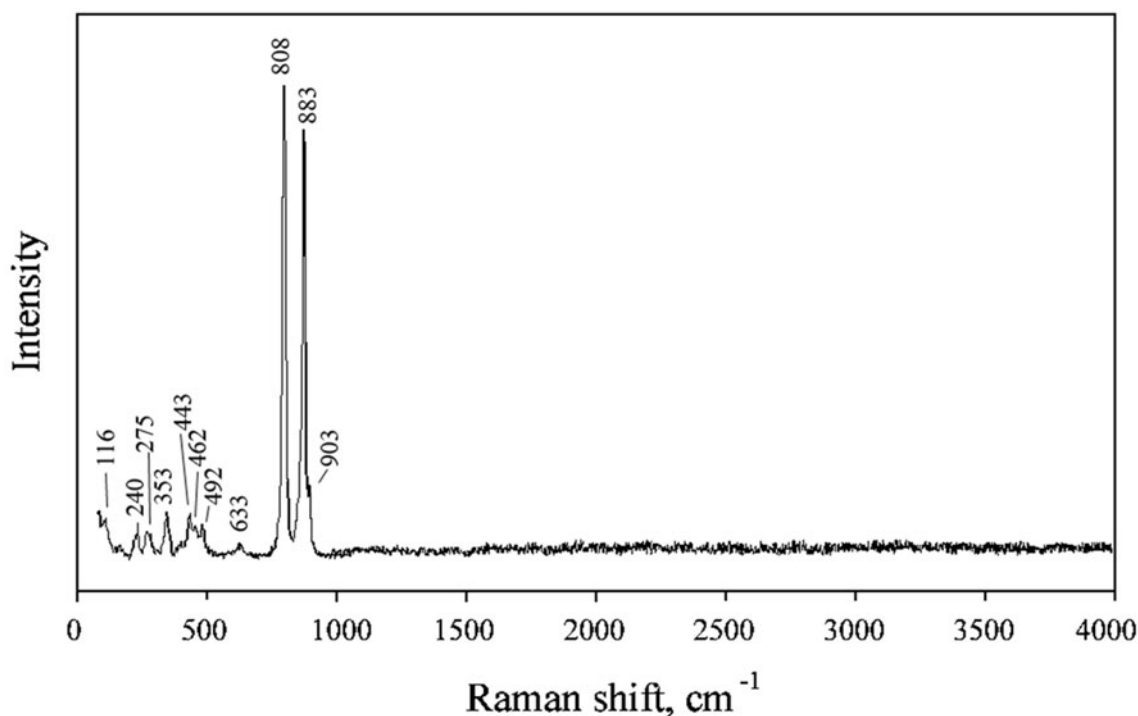
In this article we describe the new mineral evseevite (Cyrillic: *евсеевит*), ideally  $\text{Na}_2\text{Mg}(\text{AsO}_4)\text{F}$ . It is named in honour of the Russian mineralogist Aleksandr Andreevich Evseev (born 1949) who works in the Fersman Mineralogical Museum of the Russian Academy of Sciences, Moscow. He is a specialist in the history of mineralogy and the application of geographical information in mineralogy.

Both the new mineral and its name (symbol *Evs*) have been approved by the Commission on New Minerals, Nomenclature and Classification of the International Mineralogical Association

(IMA2019-064, Pekov *et al.*, 2019c). The holotype specimen is deposited in the systematic collection of the Fersman Mineralogical Museum with the catalogue number 96701.

### Occurrence and general appearance

The Arsenatnaya fumarole, discovered by us in 2012, is located at the summit of the Second scoria cone of the Northern Breakthrough of the Great Tolbachik Fissure Eruption, a monogenetic volcano formed in 1975 (Fedotov and Markhinin, 1983).



**Figure 2.** The Raman spectrum of evseevite (the holotype).

**Table 1.** Chemical composition (in wt.%) of evseevite.

Constituent	Holotype (#6328)			Cotype (#6260)			Probe standard
	Mean*	Range	S.D.	Mean*	Range	S.D.	
Na <sub>2</sub> O	26.67	26.11–27.09	0.45	28.83	28.11–29.25	0.41	albite
K <sub>2</sub> O	0.23	0.04–0.37	0.12	–	–	–	microcline
CaO	0.66	0.53–0.87	0.15	–	–	–	wollastonite
MgO	17.05	16.34–17.83	0.64	18.51	17.72–19.31	0.54	chromite
CuO	0.23	0.00–0.39	0.17	–	–	–	Cu
ZnO	0.30	0.21–0.38	0.09	–	–	–	ZnS
Fe <sub>2</sub> O <sub>3</sub> **	0.45	0.34–0.57	0.10	1.17	0.37–1.53	0.40	magnetite
SiO <sub>2</sub>	0.22	0.16–0.30	0.06	–	–	–	wollastonite
P <sub>2</sub> O <sub>5</sub>	–	–	–	8.17	6.89–9.17	0.88	LaPO <sub>4</sub>
As <sub>2</sub> O <sub>5</sub>	48.83	47.97–49.39	0.62	36.21	34.12–39.51	1.99	InAs
SO <sub>3</sub>	–	–	–	2.65	0.94–4.53	1.55	ZnS
F	8.01	7.78–8.21	0.19	6.78	6.31–7.55	0.44	fluorophlogopite
–O=F	–3.37	–	–	–2.85	–	–	–
Total	99.28	–	–	99.47	–	–	–

S.D. = standard deviation; ‘–’ indicates the content is below the detection limit; \*averaged for four spot analyses for the holotype and for seven analyses for the cotype; \*\*the trivalent state of admixed iron is assumed because of the strongly oxidising conditions of mineral formation in the Arsenatnaya fumarole: all iron minerals known from here contain only Fe<sup>3+</sup> (Pekov *et al.*, 2018a).

Arsenatnaya is one of the hottest Tolbachik fumaroles today: the temperatures measured by us using a chromel–alumel thermocouple in the period 2012–2022, reached 500°C in its deep levels. It is a locality outstanding in mineral diversity and originality: more than 200 mineral species have been reliably identified here, including 67 new minerals. The mineralogy and zonation of sublimate incrustations of Arsenatnaya were described recently by Pekov *et al.* (2018a) and Shchipalkina *et al.* (2020); the general mineralogical features of oxidising-type fumaroles of Tolbachik were over-viewed by Vergasova and Filatov (2016) and Pekov *et al.* (2020a).

The specimens with the new mineral were first collected by us in July 2018 from hot pockets situated 1.5–2 m below the day surface, within zone IV (Shchipalkina *et al.*, 2020), or polyminerals

zone. Evseevite is represented by two chemical varieties. The variety close to the end-member Na<sub>2</sub>Mg(AsO<sub>4</sub>)F gave the specimen considered as the holotype (sample<sup>1</sup> #6328): a suite of complex studies was carried out on samples of this composition, including the crystal structure determination. The variety enriched with admixed P and S, which partially substitute As, is considered as the cotype (#6260) and was subjected to scanning electron microscopy (SEM), electron microprobe analysis (EMPA) and powder X-ray diffraction (XRD). Additional specimens of the

<sup>1</sup>#6328, 6260 and 7529 are authors’ working numbers of the three samples studied; #6328 corresponds to the catalogue number 96701 in the systematic collection of the Fersman Mineralogical Museum (see Introduction).

variety chemically close to the holotype but morphologically different were collected from other pockets in the same area in July 2022 (#7529). The temperature measured in the pockets with evseevite during sampling varied from 380 to 450°C.

These chemical varieties of evseevite occur in different mineral assemblages. The holotype (and later collected sample #7529) is associated with sanidine, hematite, tenorite, aegirine, cassiterite, sylvite, halite, johillerite, badalovite, calciojohillerite, hatertite, arsmirandite, yurmarinite, axelite, polyarsite, apthitalite, potassic-magnesium-fluoro-arfvedsonite, litidionite, ferrisanidine and tridymite. The minerals associated with the cotype are hematite, fluorophlogopite, svabite, fluorapatite, tilasite, calciojohillerite, forsterite, cassiterite, belomarinaite and apthitalite.

Evseevite occurs in cavities as prismatic, typically long-prismatic to acicular or hair-like crystals up to 0.1 mm, rarely up to 0.7 mm long and up to 0.03 mm thick. They are elongated along [100] and usually combined in parallel, near-parallel, sheaf-, bush- or brush-like aggregates. Clusters (up to 0.5 mm across) of randomly oriented crystals are also common. Acicular crystals form interrupted brushes and hair-like crystals compose pilous crusts (Fig. 1) up to 2 × 2 mm in area. Some crystals are skeletal and case-like.

We suggest that evseevite was deposited directly from the gas phase as a volcanic sublimate or, more probably, formed as a result of the interaction between fumarolic gas and basalt scoria at the temperatures not lower than 450°C. Basalt could be a source of Mg which has low volatility in such fumarolic systems (Symonds and Reed, 1993).

### Physical properties and optical data

Evseevite is transparent, colourless or pale pinkish, with a white streak and vitreous lustre. The mineral is brittle, cleavage or parting was not observed. The fracture is uneven. Density calculated using the empirical formula and unit-cell volume found from powder XRD data is 3.377 g cm<sup>-3</sup> for the holotype and 3.226 g cm<sup>-3</sup> for the P- and S-enriched cotype.

The optical data were obtained for the holotype specimen. It is optically biaxial (+),  $\alpha = 1.545(2)$ ,  $\beta = 1.546(2)$ ,  $\gamma = 1.549(2)$ ,  $2V_{\text{meas}} = 40(10)^\circ$  and  $2V_{\text{calc}} = 60^\circ$  (589 nm). Dispersion of optical axes was not observed. Orientation:  $X = a$ . Extinction is straight and elongation is negative. In transmitted plane-polarised light the mineral is colourless and non-pleochroic.

### Raman spectroscopy

The Raman spectrum of the holotype evseevite (Fig. 2) was obtained on a randomly oriented crystal using an EnSpectr R532 instrument with a green laser (532 nm) at room temperature. The output power of the laser beam was ~10 mW. The spectrum was processed using the EnSpectr expert mode program in the range from 4000 to 100 cm<sup>-1</sup> with the use of a holographic diffraction grating with 1800 lines per cm<sup>-1</sup> and a resolution of 6 cm<sup>-1</sup>. The diameter of the focal spot on the sample was ~10 μm. The back-scattered Raman signal was collected with a 40× objective; signal acquisition time for a single scan of the spectral range was 1500 ms and the signal was averaged over 20 scans.

The bands in the Raman spectrum of evseevite are assigned according to Nakamoto (1986). Bands in the region between 950 and 770 cm<sup>-1</sup> correspond to As<sup>5+</sup>-O stretching vibrations of AsO<sub>4</sub><sup>3-</sup> anions. The presence of two strong bands in this region (with maxima at 883 and 808 cm<sup>-1</sup>) is caused by significant

**Table 2.** Powder X-ray diffraction data ( $d$  in Å) of evseevite (the holotype).

$l_{\text{obs}}$	$d_{\text{obs}}$	$l_{\text{calc}}^*$	$d_{\text{calc}}$	$hkl$
14	7.05	13	7.063	020
11	6.08	10	6.087	021
2	4.982	4	4.981	110
4	4.599	3	4.600	111
<b>100</b>	<b>4.001</b>	100	4.007	121
5	3.831	6	3.833	112
<b>31</b>	<b>3.527</b>	28, 6	3.531, 3.527	040, 130
<b>56</b>	<b>3.479</b>	63	3.482	023
5	3.380	4	3.384	131
5	3.117	5	3.119	113
<b>45</b>	<b>3.041</b>	44	3.044	042
<b>29</b>	<b>2.998</b>	28	3.001	004
2	2.916	1	2.914	123
3	2.851	1	2.858	141
<b>44</b>	<b>2.657</b>	51	2.661	200
<b>68</b>	<b>2.642</b>	10, 64	2.646, 2.642	133, 142
<b>36</b>	<b>2.613</b>	40	2.614	104
6	2.494	8	2.495	150
12	2.437	7, 6	2.438, 2.433	221, 202
6	2.353	5	2.354	060
4	2.283	3	2.286	134
4	2.274	2, 2	2.275, 2.273	231, 025
1	2.187	1	2.189	213
3	2.161	2	2.161	232
2	2.128	1	2.125	240
4	2.119	2, 1	2.119, 2.117	161, 153
3	2.099	2	2.101	144
5	2.091	4	2.090	125
4	2.028	4	2.029	063
<b>33</b>	<b>2.002</b>	35, 4	2.003, 2.001	242, 006
16	1.992	15	1.991	204
2	1.971	2	1.972	214
3	1.918	2	1.919	154
4	1.913	4	1.912	251
6	1.895	7	1.896	163
5	1.872	4	1.873	106
3	1.856	3	1.857	116
2	1.844	2	1.843	252
1	1.833	1	1.834	234
1	1.799	1	1.800	172
15	1.765	16	1.766	080
<b>32</b>	<b>1.741</b>	37	1.741	046
5	1.703	4	1.703	321
2	1.690	1, 2	1.692, 1.689	262, 312
4	1.666	1, 3	1.667, 1.666	235, 027
4	1.662	3, 1	1.660, 1.660	330, 181
2	1.644	3	1.645	331
5	1.613	1, 3, 3	1.615, 1.614, 1.611	083, 182, 313
4	1.602	4	1.603	165
3	1.591	1, 2	1.594, 1.590	271, 127
1	1.554	2	1.553	272
12	1.532	4, 13	1.533, 1.533	333, 342
9	1.529	6	1.527	304
5	1.524	1, 3, 1	1.525, 1.522, 1.520	066, 084, 264
3	1.502	4	1.502	350
1	1.493	1	1.491	351
6	1.471	9	1.471	280
9	1.463	12	1.463	184
11	1.457	15	1.457	246
4	1.444	6	1.444	108
1	1.433	1	1.434	217
1	1.429	1	1.429	282
4	1.412	4	1.412	227
3	1.408	1, 3	1.407, 1.407	361, 353
1	1.399	1	1.399	325

\*For the calculated pattern, only reflections with intensities  $\geq 1$  are given. The strongest reflections are marked in boldtype.

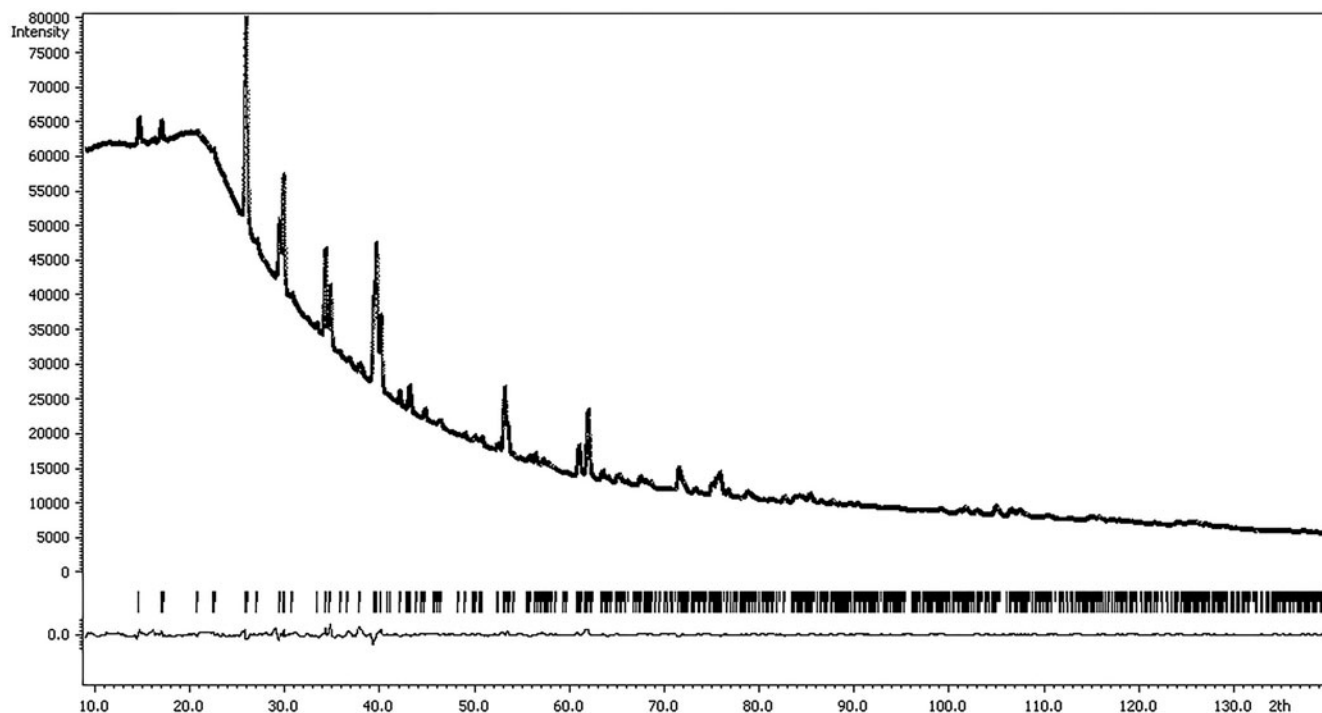
distortion of AsO<sub>4</sub> tetrahedron (see below). Bands with frequencies lower than 520 cm<sup>-1</sup> correspond to bending vibrations of AsO<sub>4</sub> tetrahedra, Mg–O stretching vibrations and lattice modes.

**Table 3.** Comparative data of evseevite (holotype and cotype) and moraskoite.

Mineral	Evseevite (holotype)	Evseevite (cotype, P- and S-enriched variety)	Moraskoite*
Ideal formula	Na <sub>2</sub> Mg(AsO <sub>4</sub> )F		Na <sub>2</sub> Mg(PO <sub>4</sub> )F
Empirical formula	(Na <sub>1.99</sub> Ca <sub>0.03</sub> K <sub>0.01</sub> ) <sub>Σ2.03</sub> (Mg <sub>0.98</sub> Fe <sub>0.01</sub> Zn <sub>0.01</sub> Cu <sub>0.01</sub> ) <sub>Σ1.01</sub> [(As <sub>0.98</sub> Si <sub>0.01</sub> ) <sub>Σ1.01</sub> O <sub>4</sub> ](F <sub>0.97</sub> O <sub>0.03</sub> )	Na <sub>2.02</sub> (Mg <sub>1.00</sub> Fe <sub>0.03</sub> ) <sub>Σ1.03</sub> [(As <sub>0.69</sub> P <sub>0.25</sub> S <sub>0.07</sub> ) <sub>Σ1.01</sub> O <sub>4</sub> ] (F <sub>0.78</sub> O <sub>0.22</sub> )	Na <sub>1.88</sub> (Mg <sub>1.06</sub> Fe <sub>0.01</sub> ) <sub>Σ1.07</sub> [P <sub>1.00</sub> O <sub>4.00</sub> ]F <sub>1.00</sub>
Crystal system	Orthorhombic		Orthorhombic
Space group	<i>Pbcn</i>		<i>Pbcn</i>
<i>a</i> (Å)	5.3224(1)	5.307(2)	5.2117(10)
<i>b</i> (Å)	14.1255(3)	14.053(3)	13.711(3)
<i>c</i> (Å)	12.0047(3)	11.940(3)	11.665(2)
<i>V</i> (Å <sup>3</sup> )	902.53(4)	890.5(8)	833.6(3)
Z	8	8	8
Strongest reflections of the powder	4.001–100	3.984–100	3.909–75
X-ray diffraction pattern:			
<i>d</i> (Å) – <i>I</i> (%)	3.527–31	3.511–26	3.382–52
	3.479–56	3.467–48	2.955–90
	3.041–45	3.027–35	2.606–100
	2.998–29	2.992–27	2.571–96
	2.657–44	2.649–95	2.547–68
	2.642–68	2.631–56	1.691–67
	2.613–36	2.608–31	
	2.002–33	1.995–32	
	1.741–32	1.733–24	
<i>D</i> <sub>calc</sub> (g cm <sup>-3</sup> )	3.337	3.226	2.925
Optical data	Biaxial (+)	none examined	Biaxial
$\alpha$	1.545(2)		
$\beta$	1.546(2)		$n_{\text{mean}} = 1.550(4)^{**}$
$\gamma$	1.549(2)		
birefringence	0.004		ca. 0.004
2 <i>V</i> <sub>meas</sub>	40(10) <sup>o</sup>		small
Source	this work	this work	Karwowski <i>et al.</i> (2015)

\*A powder X-ray diffraction study of moraskoite was not carried out, only the calculated data were reported.

\*\*It is likely that the mean refractive index for moraskoite was determined wrongly. Our calculation of the Gladstone–Dale compatibility index for moraskoite using  $n_{\text{mean}} = 1.550$  gave  $1 - (K_p/K_c) = -0.079$ , fair. The estimated value of  $n_{\text{mean}}$  for moraskoite calculated using the Gladstone–Dale equation (Mandarino, 1981) is ca. 1.51 [1.510 for  $1 - (K_p/K_c) = 0.000$ ].



**Figure 3.** Measured and calculated powder X-ray diffraction patterns of evseevite (the holotype). The solid line corresponds to calculated data, the crosses correspond to the measured pattern, vertical bars mark all possible Bragg reflections. The difference between the measured and calculated patterns is shown by curve at the bottom.

**Table 4.** Crystal data, data collection information and structure refinement details for evseevite (the holotype).

	Rietveld refinement	Single-crystal data
Ideal formula	----- Na <sub>2</sub> Mg(AsO <sub>4</sub> )F -----	
Formula weight	228.2	
Temperature (K)	293(2)	
Radiation and wavelength (Å)	CoKα; 1.79021	MoKα; 0.71073
Crystal system, space group, Z	Orthorhombic, <i>Pbcn</i> , 8	
Unit cell dimensions (Å)	<i>a</i> = 5.3224(1) <i>b</i> = 14.1255(3) <i>c</i> = 12.0047(3)	<i>a</i> = 5.3173(6) <i>b</i> = 14.087(2) <i>c</i> = 12.0110(13)
<i>V</i> (Å <sup>3</sup> )	902.53(4)	899.71(19)
Absorption coefficient μ (mm <sup>-1</sup> )	20.042	7.823
<i>F</i> <sub>000</sub>	864	
Diffractometer	Rigaku R-Axis RAPID II	Xcalibur S CCD
Range for data collection (°)	2θ from 9.00 to 140.00	θ from 2.892 to 28.277
<i>hkl</i>		-7 ≤ <i>h</i> ≤ 7, -4 ≤ <i>k</i> ≤ 18, 0 ≤ <i>l</i> ≤ 16
Final <i>R</i> indices	<i>R</i> <sub>exp</sub> = 0.0069, <i>R</i> <sub>wp</sub> = 0.0068 <i>R</i> <sub>p</sub> = 0.0047, <i>R</i> <sub>obs</sub> = 0.0435	<i>R</i> <sub>1</sub> = 0.1106, <i>wR</i> <sub>2</sub> = 0.2188 [ <i>I</i> > 2σ( <i>I</i> )] <i>R</i> <sub>1</sub> = 0.1376, <i>wR</i> <sub>2</sub> = 0.2314 (all data)
GoF	0.98	1.30
Largest diff. peak and hole (e <sup>-</sup> /Å <sup>3</sup> )	0.84 and -0.66	2.25 and -2.25
Other data		Crystal size 0.02 × 0.05 × 0.07 mm Structure solution – direct methods Refinement method – full-matrix least-squares on <i>F</i> <sup>2</sup> Absorption correction ‘Gaussian’

The absence of bands with frequencies higher than 950 cm<sup>-1</sup> indicates the absence of groups with O–H, C–H, C–O, N–H, N–O and B–O bonds in evseevite.

### Chemical composition

The chemical composition of evseevite was studied by EMPA using a Jeol JSM-6480LV scanning electron microscope equipped with an INCA-Wave 500 wavelength-dispersive spectrometer (Laboratory of Analytical Techniques of High Spatial Resolution, Dept. of Petrology, Moscow State University), with an acceleration voltage of 20 kV, a beam current of 20 nA, and a 3 μm beam diameter.

The chemical data in wt.% are given in Table 1. Contents of other elements with atomic numbers higher than carbon were below detection limits.

The empirical formulae calculated on the basis of O+F = 5 atoms per formula unit are (Na<sub>1.99</sub>Ca<sub>0.03</sub>K<sub>0.01</sub>)Σ<sub>2.03</sub>(Mg<sub>0.98</sub>Fe<sub>0.01</sub>Zn<sub>0.01</sub>Cu<sub>0.01</sub>)Σ<sub>1.01</sub>[(As<sub>0.98</sub>Si<sub>0.01</sub>)Σ<sub>1.01</sub>O<sub>4</sub>](F<sub>0.97</sub>O<sub>0.03</sub>) for the holotype and Na<sub>2.02</sub>(Mg<sub>1.00</sub>Fe<sub>0.03</sub>)Σ<sub>1.03</sub>[(As<sub>0.69</sub>P<sub>0.25</sub>S<sub>0.07</sub>)Σ<sub>1.01</sub>O<sub>4</sub>](F<sub>0.78</sub>O<sub>0.22</sub>) for the cotype. The idealised formula is Na<sub>2</sub>Mg(AsO<sub>4</sub>)F which requires Na<sub>2</sub>O 27.16, MgO 17.66, As<sub>2</sub>O<sub>5</sub> 50.36, F 8.33, -O=F – 3.51, total 100 wt.%.

The Gladstone–Dale compatibility index (Mandarino, 1981) for the holotype evseevite, 1 – (*K<sub>p</sub>*/*K<sub>c</sub>*) = 0.028, excellent.

### X-ray crystallography and crystal structure determination details

Single-crystal XRD studies of the holotype sample of evseevite were carried out at room temperature using an Xcalibur S diffractometer equipped with a CCD detector (MoKα-radiation). Powder XRD data for both holotype and cotype samples were collected with a Rigaku R-AXIS Rapid II single-crystal diffractometer equipped with cylindrical image plate detector (radius 127.4 mm) using Debye-Scherrer geometry, CoKα radiation (rotating anode with VariMAX microfoc optics), 40 kV, 15 mA and 15 min exposure. Angular resolution of the detector is 0.045°2θ (pixel size 0.1 mm).

**Table 5.** Coordinates and isotropic displacement parameters (*U*<sub>iso</sub>, in Å<sup>2</sup>) of atoms for evseevite (the holotype).

Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> <sub>iso</sub>
As	0.70138(19)	0.88053(11)	0.58705(8)	0.0245(3)
Mg	0.7307(4)	0.51258(18)	0.8186(2)	0.0343(15)
Na(1)	0.7548(7)	0.7389(3)	0.8311(3)	0.0080(13)
Na(2)	0.7413(7)	0.6223(4)	0.5832(2)	0.0308(9)
O(1)	0.8051(9)	0.77571(18)	0.6306(4)	0.0147(6)
O(2)	0.8633(5)	0.9662(2)	0.6536(3)	0.0147(6)
O(3)	0.7442(10)	0.8930(3)	0.45019(13)	0.0147(6)
O(4)	0.3846(3)	0.8969(2)	0.6075(3)	0.0147(6)
F(1)	½	0.6201(3)	¾	0.0147(6)
F(2)	0	0.5923(3)	¾	0.0147(6)

The data were integrated using the software package *Osc2Tab* (Britvin et al., 2017). Powder XRD data (in Å for CoKα) for the holotype are given in Table 2. The strongest reflections of powder XRD patterns and unit-cell parameters calculated from powder data for both holotype and cotype are presented in Table 3.

The single-crystal XRD dataset used for the structure model determination was obtained on the same diffractometer Xcalibur S CCD. Data reduction was performed using *CrysAlisPro*, version 1.171.37.34 (Agilent Technologies, 2014). The data were corrected for Lorentz factor and polarisation effects. The crystal structure was solved by direct methods and refined in the space group *Pbcn* using the *SHELX* software package (Sheldrick, 2015). The low quality of the crystal and consequently of the experimental data precluded obtaining an excellent agreement between observed and calculated *F* values but resulted in an acceptable agreement with *R*<sub>hkl</sub> = 0.1106 for 1116 reflections with *I* > 2σ(*I*). Reasonable values of interatomic distances and displacement parameters of atoms, as well as good agreement between the measured and calculated powder XRD patterns (Table 2; Fig. 3) showed that the obtained structure model is correct. Further refinement of the structure of holotype evseevite was performed by the Rietveld method using this model.

**Table 6.** Selected interatomic distances (Å) in the structure of evseevite (the holotype).

As–O(1)	1.665(3)	Na(1)–O(1)	2.477(6)
As–O(2)	1.687(3)	Na(1)–O(1)	2.443(6)
As–O(3)	1.668(2)	Na(1)–O(3)	2.349(5)
As–O(4)	1.7194(19)	Na(1)–O(4)	2.465(5)
<As–O>	1.685	Na(1)–F(1)	2.367(5)
		Na(1)–F(2)	2.634(5)
Mg–O(2)	2.089(4)	<Na1–(O,F)>	2.456
Mg–O(2)	2.145(4)		
Mg–O(3)	2.071(4)	Na(2)–O(1)	2.266(6)
Mg–O(4)	2.031(4)	Na(2)–O(2)	2.426(6)
Mg–F(1)	2.120(4)	Na(2)–O(3)	2.685(7)
Mg–F(2)	2.000(3)	Na(2)–O(3)	2.715(7)
<Mg–(O,F)>	<2.076>	Na(2)–O(4)	2.429(5)
		Na(2)–F(1)	2.379(3)
		Na(2)–F(2)	2.467(3)
		<Na2–(O,F)>	2.894

**Table 7.** Bond valence calculations for evseevite (the holotype).

	As	Mg	Na(1)	Na(2)	Σ
O(1)	1.33		0.16 0.17	0.26	1.92
O(2)	1.25	0.34 0.30		0.18	2.07
O(3)	1.32	0.35	0.21	0.09 0.09	2.06
O(4)	1.14	0.38	0.16	0.17	1.85
F(1)		0.23 <sup>×2→</sup>	0.15 <sup>×2→</sup>	0.15 <sup>×2→</sup>	1.06
F(2)		0.32 <sup>×2→</sup>	0.08 <sup>×2→</sup>	0.12 <sup>×2→</sup>	1.04
Σ	5.04	1.92	0.93	1.06	

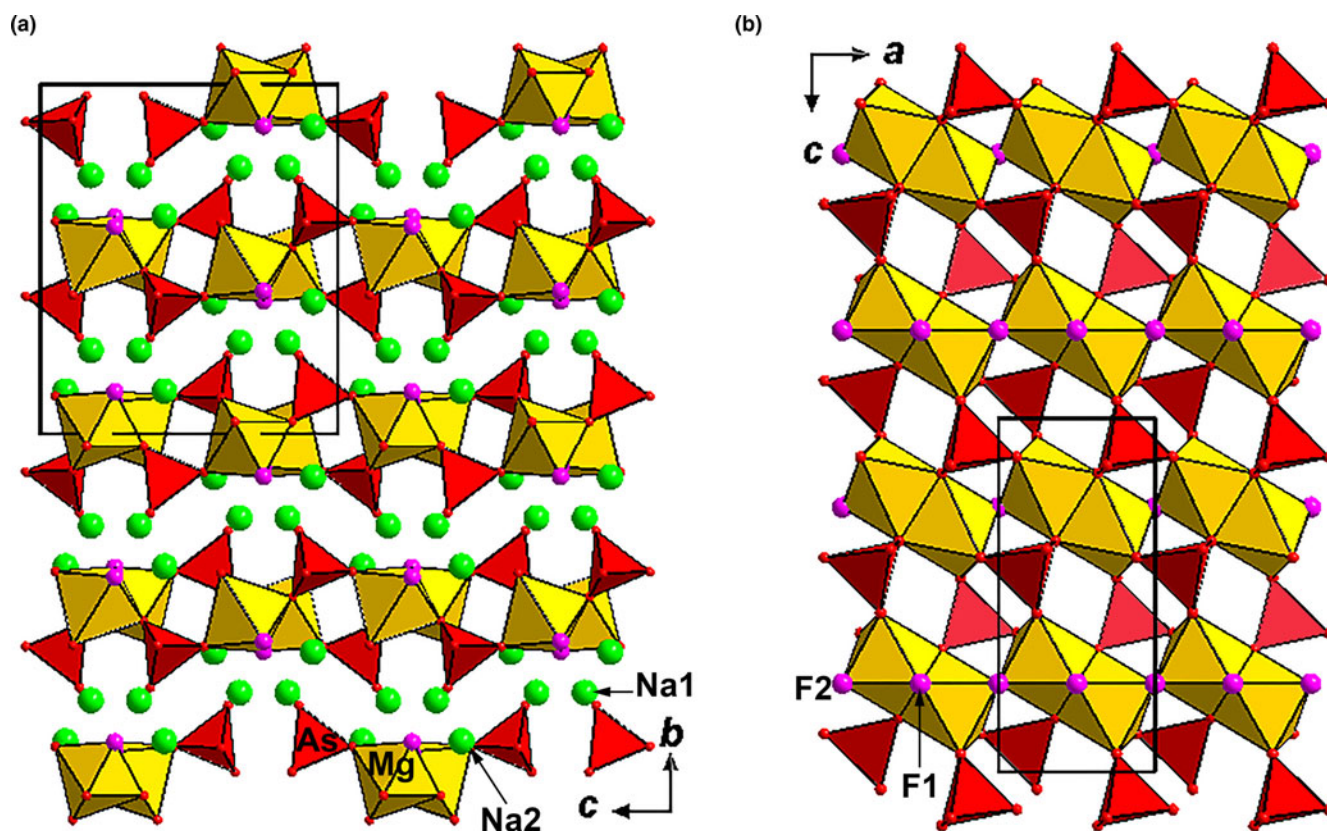
Bond-valence parameters for As–O, Mg–O and Na–O are taken from Gagné and Hawthorne (2015), and for Mg–F and Na–F from Brese and O’Keeffe (1991).

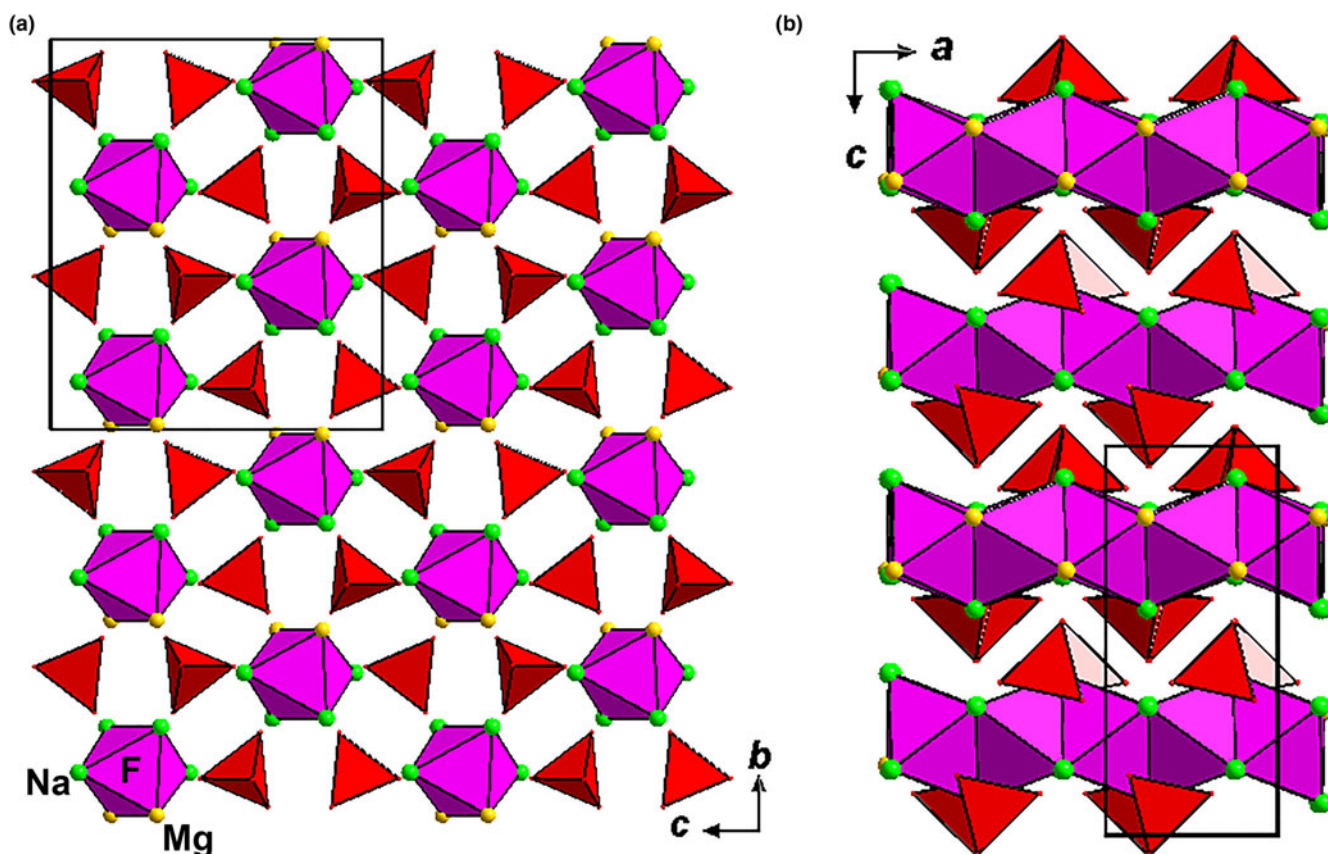
Data treatment and the Rietveld structure analysis were carried out using the *JANA2006* program package (Petříček *et al.*, 2006). The profiles were modelled using a pseudo-Voigt function. The structure was refined in isotropic approximation of atomic thermal displacements, the values of  $U_{\text{iso}}$  for all anions were restricted to be equal. The interatomic distances for As-centred tetrahedron and Mg-centred octahedron were softly restrained nearby the values obtained for the single-crystal model. Final agreement factors are:  $R_{\text{wp}} = 0.0068$ ,  $R_{\text{p}} = 0.0047$ ,  $R_{\text{obs}} = 0.0435$ . The observed and calculated powder XRD diagrams demonstrate very good agreement (Fig. 3; Table 2).

Data collection information and structure refinement details for both single-crystal and powder XRD studies are presented in Table 4, coordinates and thermal displacement parameters of atoms are given in Table 5, selected interatomic distances in Table 6, and bond valence calculations in Table 7. The data presented in Tables 5–7 are obtained in the result of the Rietveld refinement. The crystallographic information file has been deposited with the Principal Editor of *Mineralogical Magazine* and is available as Supplementary material (see below).

## Discussion

Evseevite, ideally  $\text{Na}_2\text{Mg}(\text{AsO}_4)\text{F}$ , is the isostructural arsenate analogue of moraskoite  $\text{Na}_2\text{Mg}(\text{PO}_4)\text{F}$  (Karwowski *et al.*, 2015) and synthetic  $\text{Na}_2\text{Mg}(\text{PO}_4)\text{F}$  (Swafford and Holt, 2002). The structure of evseevite (Fig. 4a) is based upon the heteropolyhedral (010) layers formed by [100] chains of Mg-centred octahedra connected via  $\text{AsO}_4$  tetrahedra (Fig. 4b). A repeat unit of the chain is dimer  $[\text{Mg}_2\text{O}_6\text{F}_2]^{10-}$  consisting of two  $\text{MgO}_4\text{F}_2$  octahedra sharing

**Figure 4.** The crystal structure of evseevite (a) projected along the *a* axis and (b) the heteropolyhedral layer in it. The unit cell is outlined.



**Figure 5.** F-centred octahedra  $\text{FNa}_4\text{Mg}_2$  and  $\text{AsO}_4$  tetrahedra in the structure of evseevite (two projections). The unit cell is outlined.

a common face of  $\text{O}(2)\text{--F}(1)\text{--O}(2)$ . Adjacent dimers are linked *via*  $\text{F}(2)$  vertices. Na cations occupy two crystallographically non-equivalent sites.  $\text{Na}(1)$  cations occupy centre  $\text{Na}(1)\text{O}_4\text{F}_2$  octahedra [in moraskoite this site centres seven-fold polyhedron with one elongated  $\text{Na}(1)\text{--O}(1)$  distance of 2.93 Å; in evseevite this distance is > 3.0 Å and, thus,  $\text{O}(1)$  is excluded from the coordination sphere of the  $\text{Na}(1)$  cation].  $\text{Na}(2)$  cations occupy seven-fold polyhedra  $\text{Na}(2)\text{O}_5\text{F}_2$ . The  $\text{Na}(1)$  sites are located between the heteropolyhedral layers whereas the  $\text{Na}(2)$  sites are situated at both sides of the heteropolyhedral layer.

The crystal structure of evseevite, as well as of moraskoite, can be described in terms of anion-centred polyhedra. Both  $\text{F}(1)$  and  $\text{F}(2)$  sites are octahedrally coordinated by two Mg and four Na cations each. F-centred octahedra  $[\text{FNa}_4\text{Mg}_2]^{7+}$  share faces to form [100] chains  $[\text{FNa}_2\text{Mg}]^{3+}$ ;  $\text{AsO}_4$  tetrahedra are located between the chains (Fig. 5). This approach was used for the description of nacaphite  $\text{Na}_2\text{Ca}(\text{PO}_4)\text{F}$  (Krivovichev *et al.*, 2007), the monoclinic ( $P2_1/c$ ) mineral which is structurally related to moraskoite and evseevite and has the same stoichiometry. The relation of the crystal structures and unit-cell metrics of nacaphite and moraskoite were reported in detail by Karwowski *et al.* (2015).

Evseevite, moraskoite and nacaphite belong to a group of fairly few minerals with antiperovskite structure, i.e. they have perovskite-type structures but with anions replaced by cations and *vice versa*. The group of natural antiperovskites is known to include sulfates, phosphates and silicates (Krivovichev, 2008; Karwowski *et al.*, 2015), and now with the addition of evseevite has the first arsenate mineral with antiperovskite units: its

structure is based on F-centred octahedra  $[\text{FNa}_4\text{Mg}_2]^{7+}$  which form chains similar to those in hexagonal  $2H$ -perovskites. We have not found any information on synthetic arsenate antiperovskites in literature and databases, however, the X-centred octahedra  $[\text{XNa}_6]^{5+}$  ( $X = \text{F}, \text{OH}$ ) are present in the structures of synthetic cubic arsenates  $\text{Na}_7\text{F}(\text{AsO}_4)_2 \cdot 19\text{H}_2\text{O}$  and  $\text{Na}_7(\text{OH})(\text{AsO}_4)_2 \cdot 19\text{H}_2\text{O}$  (Baur and Tillmanns, 1974) isostructural to the mineral natrophosphate, ideally  $\text{Na}_7\text{F}(\text{PO}_4)_2 \cdot 19\text{H}_2\text{O}$ . Avdontceva *et al.* (2021) suggested that the presence of the  $[\text{FNa}_6]^{5+}$  units makes it possible to consider natrophosphate-type compounds as a precursor for the formation of antiperovskite structure motifs based upon anion-centred octahedra.

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**Supplementary material.** The supplementary material for this article can be found at <https://doi.org/10.1180/mgm.2023.50>.

**Competing interests.** The authors declare none.

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