

## Microscopy and Chemical Analysis of Topological Insulator $\text{Bi}_2\text{Se}_3$ and Topological Crystalline Insulator $\text{SnTe}$ Nanostructures

Judy J. Cha<sup>1,2</sup>

<sup>1</sup> Department of Mechanical Engineering & Materials Science, Yale University, New Haven, CT, USA

<sup>2</sup> Energy Sciences Institute, Yale West Campus, West Haven, CT, USA

Topological insulators and topological crystalline insulators are a new quantum matter whose band structure belongs to a different topology compared to a regular, trivial insulator [1, 2]. The physical consequence of this is manifestation of conducting surface or edge states that are spin-polarized and Dirac-dispersive. The unique electronic property of the surface states opens opportunities to study previously inaccessible fundamental condensed matter phenomena such as Majorana fermions [3] and magnetic monopoles [4] and to design future spin-based device applications. A plethora of experiments have been carried out to investigate the exotic electronic properties of the surface states in topological insulator  $\text{Bi}_2\text{Se}_3$  and topological crystalline insulator  $\text{SnTe}$ .

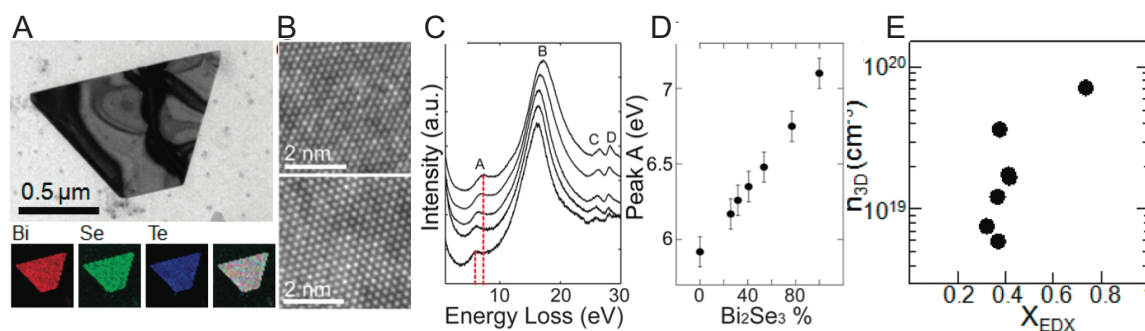
Nanostructured  $\text{Bi}_2\text{Se}_3$  and  $\text{SnTe}$  are ideal to study and exploit the properties of the surface states due to the large surface-to-volume ratios, which enhance the surface effects. In addition, the nanostructure morphology can play an important role in confining the surface electrons to follow well-defined paths for interference experiments. Magnetic and charge-compensation doping has also shown to be important for controlling the surface state property. This highlights the critical need to carefully characterize the morphology, atomic structure, and chemical composition of topological insulator and topological crystalline insulator nanostructures.

In this talk, I will present structural and chemical characterizations of  $\text{Bi}_2\text{Se}_3$  nanoribbons and  $\text{SnTe}$  nanoplates using scanning transmission electron microscopy and showcase how their structures determine their electronic properties. For chemical analysis, X-ray energy dispersive spectroscopy (EDX) was proven very useful due to the high atomic number elements such as Bi and Te that make up topological insulators and topological crystalline insulators. Monochromated electron energy-loss spectroscopy (EELS) was used to probe plasmonic and optical properties of these materials in thin platelet forms.

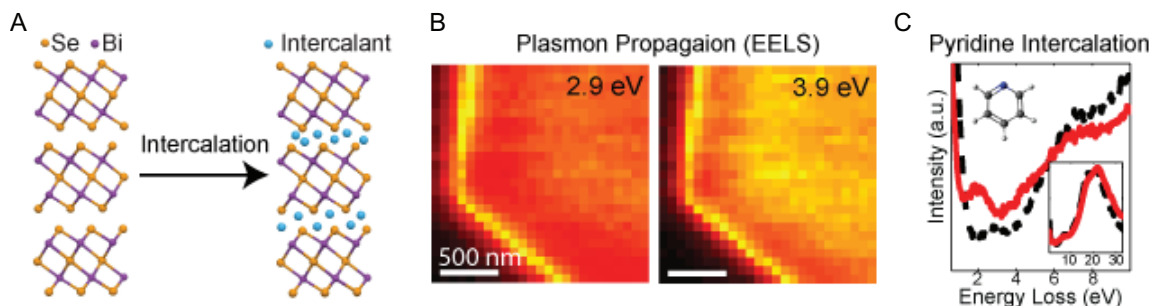
A few examples I will discuss are 1) analysis of chemical compositions of  $\text{Bi}_2(\text{Se}_x\text{Te}_{1-x})_3$  nanoribbons and nanoplates and resulting changes in the lattice constant, plasmon excitations, and electron carrier density (Fig. 1) [5], 2) Electron energy-loss spectroscopy investigation of plasmonic and optical property changes due to dielectric molecule intercalation into  $\text{Bi}_2\text{Se}_3$  nanoribbons (Fig. 2) [6-8], and 3) characterization of Indium doping concentrations in  $\text{SnTe}$  nanostructures, which induces superconductivity (Fig. 3) [9, 10].

The structure-property relation of topological insulator and topological crystalline insulator nanostructures, elucidated by analytical electron microscopy studies, provides an essential guide to improve the material quality for enhanced electrical transport properties of these materials.

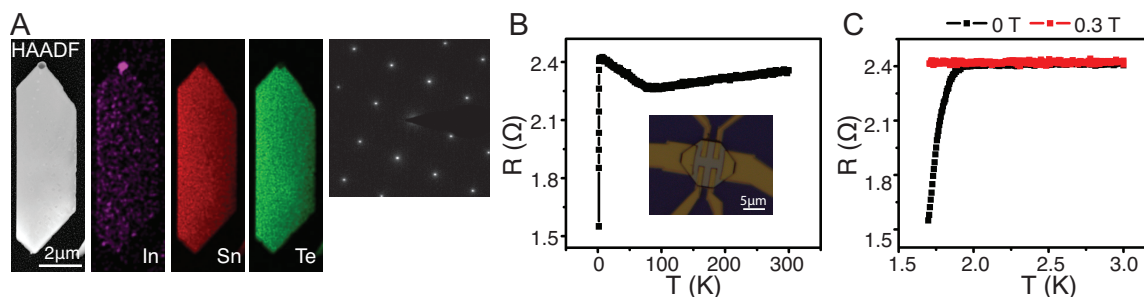
- [1] X.-L. Qi and S.-C. Zhang, *Reviews of Modern Physics* **83**, 1057 (2011).  
 [2] L. Fu, *Phys. Rev. Lett.* **106**, 106802 (2011).  
 [3] F. Wilczek, *Nature Physics* **5**, 614 (2009).  
 [4] X.-L. Qi, R. Li, J. Zang, and S.-C. Zhang, *Science* **323**, 1184 (2009).  
 [5] J. J. Cha, D. Kong, S.-S. Hong, J. G. Analytis, K. Lai, and Y. Cui, *Nano Letters* **12**, 1107 (2012).  
 [6] K. J. Koski *et al*, *J. Am. Chem. Soc.* **134**, 7584 (2012).  
 [7] J. J. Cha *et al*, *Nano Letters* **13**, 5913 (2013).  
 [8] J. Yao *et al*, *Nature Communications* **5**, 5670 (2014).  
 [9] J. Shen, Y. Jung, A. S. Disa, F. J. Walker, C. H. Ahn, and J. J. Cha, *Nano Lett.* **14**, 4183 (2014).  
 [10] J. Shen, Y. Xin, and J. J. Cha, arXiv:1410.4244 (2014).



**Figure 1.** (A) Chemical analysis of  $\text{Bi}_2(\text{Se}_x\text{Te}_{1-x})_3$  nanoplates by EDX. (B) TEM image of  $\text{Bi}_2\text{Se}_3$  (top) and  $\text{Bi}_2\text{Te}_3$  (bottom). (C) EELS spectra showing the surface plasmons of  $\text{Bi}_2(\text{Se}_x\text{Te}_{1-x})_3$  nanoplates. (D) Surface plasmon peak linearly scales with the Se/Te ratio. (E) Carrier density scales with the Se/Te ratio.



**Figure 2.** (A) Intercalation into the van der Waals gap of  $\text{Bi}_2\text{Se}_3$ . (B) Plasmon propagation in a  $\text{Bi}_2\text{Se}_3$  nanoplate by EELS mapping. (C) Optical property change by pyridine intercalation into  $\text{Bi}_2\text{Se}_3$ .



**Figure 3.** (A) Chemical analysis of In-doped  $\text{SnTe}$  by EDX. (B) Superconductivity induced by the In doping. (C) Field-cooled resistivity measurement confirms the superconductivity.