

The Effect of Oxygen on the Electronic Structure of MgB₂

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The discovery of superconductivity in MgB₂, with a transition temperature of T_c=39K [1], has focused scientific attention towards an understanding of its superconducting properties. Superconductivity in MgB₂ is driven by hole transport through the boron orbitals [2]. The presence of oxygen as segregates in the grain boundaries and precipitates in the bulk of polycrystalline materials [3] could have a large effect on the hole carrier concentration and therefore change the superconducting properties of MgB₂. In this study, we elaborate the first experimental evidence that oxygen segregates in the bulk of MgB₂ can have an effect on the hole carrier concentration.

The experiment was performed using Z-contrast imaging and electron energy loss spectroscopy (EELS) techniques. The Z-contrast images and EELS were obtained from the 200 kV JEOL 2010F STEM at the University of Illinois at Chicago, which has an objective aperture ~15 mrad, and an angular range of 0-52 mrad for EELS and 52-200 mrad for the annular dark field (ADF) detector. The spectra were acquired with an energy resolution of 1.2 eV and energy dispersion of 0.2 eV.

Figure 1a shows a low-magnification image of a typical oxygen precipitate observed in these samples (dark spot, low contrast region). The size of the grain is about 2 μm. Figure 1b shows the same region magnified. Here, the hexagonal shape of the precipitate can be distinguished, with a diameter of about 60 nm. EEL spectra of the boron K-edge and oxygen K-edge were taken from the precipitate and the region around it (which also contains oxygen). Both spectra were background subtracted and corrected for multiple-scattering contributions. Figure 2a shows the boron K-edge spectrum for the precipitate (i), the region around the precipitate (ii) and a spectrum obtained for pure MgB₂ (iii) displayed for comparison. The fine structure of spectrum ii shows clear differences from the precipitate and the pure MgB₂. The pre-peak intensity (labeled a) in spectrum ii shows a sharper feature than the precipitate and the pure MgB₂ spectra with a decrease in its intensity at 191 eV. The peaks (labeled b and c) are slightly higher and more pronounced for spectrum ii than the precipitate spectra and both are higher than the pure MgB₂ spectrum. The oxygen spectra (figure 2b) for the two regions look completely different, and this can be taken as evidence of two different phases.

Although the atomic ratios calculated from the two different regions do not differ by more than 5%, the change of intensity shown in figure 1b with the differences shown in the oxygen K-edge spectra (figure 2b) indicates the presence of a different phase. More importantly, the region around the precipitate shows a drastic change in the boron fine structure, sharpness in the pre-peak and at 195 eV of the boron K-edge (figure 2a). This pre-peak has been the focus of diverse studies [4], and an increase in its intensity is associated with a better performance of MgB₂ as a superconductor with a higher critical temperature. Work is underway to evaluate the origin of the change in the pre-peak shape and its effect on superconducting properties [5].

References:

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 [3] Klie R. F., Idrobo J. C., and Browning N. D. et al. Applied Physics Letters 80 (2002) 21.
 [4] Klie R. F. et al, cond-mat/0211295, (2002). Submitted to Phys. Rev. B
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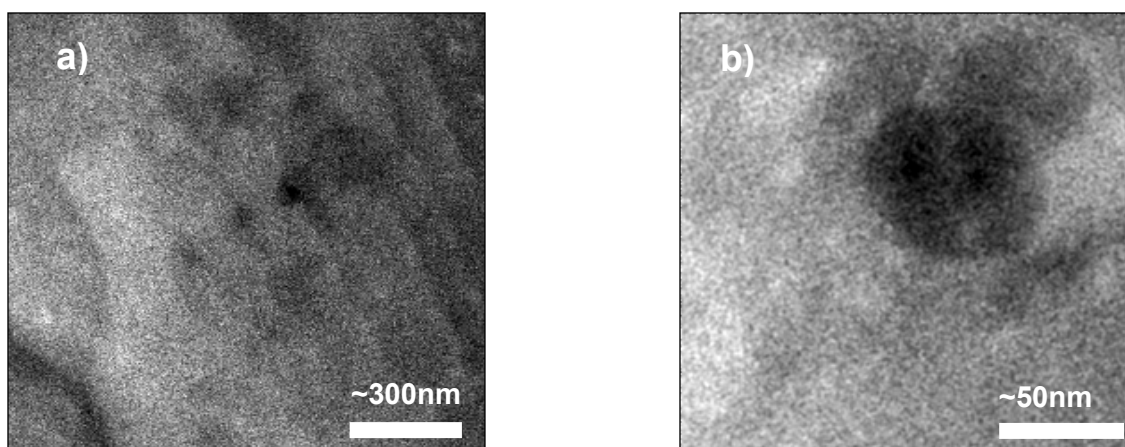


Figure 1. (a) Low magnification of polycrystalline MgB₂. (b) High magnification of figure 1a, the change of contrast in the images indicates the differences phases in the material, bulk and precipitate.

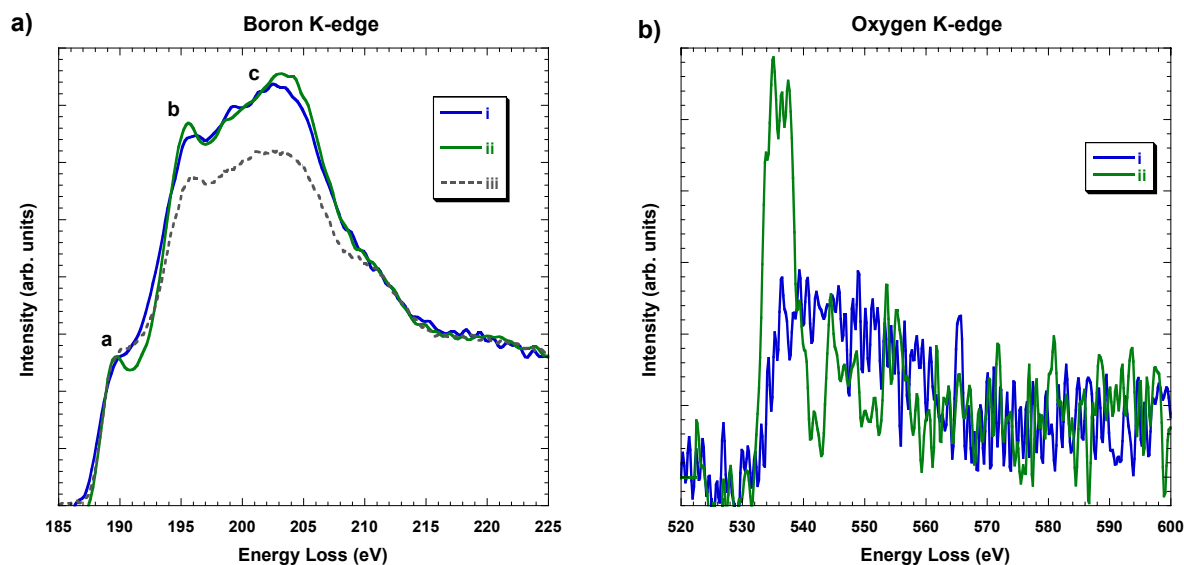


Figure 2. (a) Boron K-edge spectra, where differences can be observed in the fine structure. The changes of shape between the spectra in the pre-peak marked as *a* and peak marked as *b* are due to oxygen segregation in the bulk. (b) Oxygen K-edge spectra. The difference in the spectra for the two regions shows that oxygen has precipitated in MgB₂ forming two different phases.