

Absorption and metabolism of bioactive molecules after oral consumption of cooked edible heads of *Cynara scolymus* L. (cultivar Violetto di Provenza) in human subjects: a pilot study

E. Azzini¹, R. Bugianesi¹, F. Romano¹, D. Di Venere², S. Miccadei³, A. Durazzo¹, M. S. Foddai¹, G. Catasta¹, V. Linsalata² and G. Maiani^{1*}

¹Istituto Nazionale di Ricerca per gli Alimenti e la Nutrizione, Via Ardeatina 546, 00178 Roma, Italy

²CNR - Istituto di Scienze delle Produzioni Alimentari (ISPA), Via Amendola 122/O, 70126 Bari, Italy

³Istituto Regina Elena, CRS, Via delle Messi d'Oro 156, 00158 Roma, Italy

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The current growing interest for natural antioxidants has led to a renewed scientific attention for artichoke, due not only to its nutritional value, but, overall, to its polyphenolic content, showing strong antioxidant properties. The major constituents of artichoke extracts are hydroxycinnamic acids such as chlorogenic acid, dicaffeoylquinic acids caffeic acid and ferulic acid, and flavonoids such as luteolin and apigenin glycosides. *In vitro* studies, using cultured rat hepatocytes, have shown its hepatoprotective functions and *in vivo* studies have shown the inhibition of cholesterol biosynthesis in human subjects. Several studies have shown the effect on animal models of artichoke extracts, while information on human bioavailability and metabolism of hydroxycinnamates derivatives is still lacking. Results showed a plasma maximum concentration of 6.4 (SD 1.8) ng/ml for chlorogenic acid after 1 h and its disappearance within 2 h ($P < 0.05$). Peak plasma concentrations of 19.5 (SD 6.9) ng/ml for total caffeic acid were reached within 1 h, while ferulic acid plasma concentrations showed a biphasic profile with 6.4 (SD 1.5) ng/ml and 8.4 (SD 4.6) ng/ml within 1 h and after 8 h respectively. We observed a significant increase of dihydrocaffeic acid and dihydroferulic acid total levels after 8 h ($P < 0.05$). No circulating plasma levels of luteolin and apigenin were present. Our study confirms the bioavailability of metabolites of hydroxycinnamic acids after ingestion of cooked edible *Cynara scolymus* L. (cultivar Violetto di Provenza).

Cooked artichoke heads: Absorption: Metabolism: Hydroxycinnamic acids: Human subjects

Artichoke (*Cynara scolymus* L.) is an ancient herbaceous perennial plant, originating from the Mediterranean area, which today is cultivated all over the world. The leaves of artichoke have been widely used in herbal medicine as a choleric and diuretic since ancient times (Bruneton, 1995).

Several studies have demonstrated the efficiency and safety of artichoke extracts in the treatment of hepato-biliary dysfunction and digestive complaints in animals (Adzet *et al.* 1987; Speroni *et al.* 2003) and in human subjects (Kraft, 1997; Kirchoff *et al.* 1994). Several clinical investigations showed the ability of artichoke extract to prevent the oxidative modification of blood lipoproteins and to reduce blood cholesterol levels (Kirchoff *et al.* 1994; Gebhardt, 1998, 2002; Zapolska-Downar *et al.* 2002; Shimoda *et al.* 2003). Moreover leaf extracts have been reported to show antioxidative and protective properties against hydroperoxide-induced oxidative stress in cultured rat hepatocytes (Gebhardt, 1997; Miccadei *et al.* 2004). Artichoke beneficial effects seem due to the biochemical constituents, mainly polyphenols (chlorogenic acid (CGA), mono- and

di-caffeoylquinic acids and flavonoids such as luteolin and apigenin glycosides), sugars and inulin. The health-promoting prebiotic effects of artichoke inulin were related to the stimulation of bifidobacterial growth in the intestine (Roberfroid *et al.* 1998; Lopez-Molina *et al.* 2005). The characterization and quantification of bioactive molecules in artichoke have been studied extensively (Wang *et al.* 2003; Schütz *et al.* 2004; Zhu *et al.* 2004) while little is known on the bioavailability and bioactivity of these substance from natural or supplemental sources. Polyphenols are abundant micronutrients in our diet and evidence for their role in the prevention of degenerative disease is emerging. Bioavailability greatly differs from one polyphenol to another, so that the most abundant polyphenols in our diet are not necessarily those leading to the highest concentration of active metabolites on target tissues (Manach *et al.* 2004, 2005).

The aim of this study was to evaluate the absorption and metabolism of bioactive molecules after oral consumption of cooked edible portion of artichoke heads in human subjects.

Abbreviations: CGA, chlorogenic acid; CA, caffeic acid; DHCA, dihydrocaffeic acid; FA, ferulic acid; DHFA, dihydroferulic acid.

* **Corresponding author:** Dr. Giuseppe Maiani, fax +39 06 51494550, email maiani@inran.it

Materials and methods

Chemicals and reagents

All solvent reagents were HPLC or Optima grade; common reagents were purchased from Sigma (St Louis, MO, USA), Carlo Erba (Milan, Italy) and BDH Laboratory Supplies. (Poole, Dorset, UK) CGA, caffeic acid (CA), ferulic acid (FA), apigenin (Lyon Nord, France) and luteolin were purchased from Sigma, dihydrocaffeic acid (DHCA) from Extrasynthese (Lyon Nord, France) and dihydroferulic acid (DHFA) from AVO-CADO Research Chemicals Ltd (Morecambe, Lancashire, UK).

Subjects

Five healthy non smoking subjects (two males and three females), aged 25–32 years, body weight 61.0 (SD 9.5) kg and BMI 21.5 (SD 1.5) kg/m², were recruited on the basis of their normal blood values and clinical anamnesis from the National Institute for Research on Food and Nutrition staff (INRAN, Rome).

Each subject signed an informed consent approved by the Ethical Committee of INRAN. Several studies have shown significant effects of treatment with a small number of subjects (Hollman *et al.* 1996; McAnlis *et al.* 1999; Cremin *et al.* 2001; Rechner *et al.* 2001a; Olthof *et al.* 2001; Bugianesi *et al.* 2004).

Subjects did not use any drug or vitamin supplementation. During the three days before the experimental day, subjects were asked to follow a controlled diet limiting fruit and vegetables consumption to not more than 100 g/day, and avoiding artichokes, carrots, tomatoes, apples, pears, peaches, cherries, strawberries, blueberries, tea, coffee, wine, beer, chocolate and all their derived products. Food consumption during the three days of wash out was surveyed by a dietitian to check the diet balance by a record and recall method.

Study design

After fasting overnight, baseline blood samples were collected from each subject. After blood collection, all subjects consumed 61.7 (SD 4.4) g edible artichoke (cultivar Violetto di Provenza), steam cooked with 5.5 (SD 0.4) g extra virgin olive oil and venous blood sample collection was repeated at 0.5, 1, 2, 4, 6 and 8 h from the experimental test meal consumption.

Total energy and macronutrient content were calculated by using Food Composition Tables (National Institute for Research on Food and Nutrition, 2000). CGA, mono- and di-caffeoylquinic acids, cynarin (1,3-dicaffeoylquinic acid), luteolin-glycoside and apigenin-glycoside contents of the experimental meal are reported in Table 1.

The edible portion of artichoke heads was chosen on the basis of analysed polyphenols content of cultivar selected for the study. Artichoke heads, consumed by all participants, were collected raw and cooked in order to evaluate the polyphenol contents. Table 2 shows replicate analyses of artichoke heads both raw, and cooked as mg/100 g edible portion.

Artichoke analysis

About 25 g raw and cooked heads of artichoke were homogenized in a 250 ml flask by adding 150 ml MeOH and were

Table 1. Polyphenol contents of the experimental meal.

(Mean values and standard deviations for replicate analyses on cooked artichoke)

Polyphenol content (mg)	Mean	SD
Monocaffeoylquinic acids	41.2	0.4
Chlorogenic acid	189.7	6.3
Cynarin	11.3	0.3
Dicaffeoylquinic acids	244.5	5.2
Luteolin-glycoside	4.9	0.2
Apigenin-glycoside	6.0	0.5

extracted by stirring, heating and refluxing for 90 min with regular swirling.

The solutions obtained were filtered and the residues were extracted twice after adding 100 ml MeOH by stirring, heating and refluxing for 30 min with regular swirling. Pooled extracts were evaporated by rotavapor and reconstituted by adding 100 ml H₂O–MeOH (1:1, v/v).

The determination of polyphenols in artichoke heads was carried out in agreement with Di Venere *et al.* (2005) and the quantification was performed by HPLC using a liquid chromatograph equipped with binary gradient pump and spectrophotometric photodiode array detector (Agilent Technologies S.P.A., Milan, Italy). The Hewlett Packard Chem Station (Rev. A. 06.03) software was used for spectra and data processing. An analytical Phenomenex (Torrance, CA, USA) Ultracarb 5 ODS (30) column (4.6 × 250 mm) was used for peak separation and the elution profile was reported by Lattanzio & Van Sumere (1987). The method recovery performance was assayed by addition of suitable amounts of standards prior to the extraction of the solid samples. Recovery rates of the method were found to be >85% for caffeoylquinic acids as well as luteolin and apigenin glycosides.

The method precision was assayed by three series of extractions, each performed in triplicate using one different artichoke head. The percentage relative standard deviation for the normalized peak areas of all evaluated compounds was found to be <5% (data not shown).

Plasma samples

Blood samples were drawn into vacutainers containing EDTA as anticoagulant. Blood samples were centrifuged at 3600 g for 10 min at 4°C, plasma was separated and stored at –80°C

Table 2. Selected polyphenols analysis on raw and cooked *Cynara scolymus* L (cultivar Violetto di Provenza) heads.

(Mean values and standard deviations)

Polyphenol (mg/100 g edible portion)	Raw		Cooked	
	Mean	SD	Mean	SD
Chlorogenic acid	276.1	30.9	307.5	10.3
Total monocaffeoylquinic acids	16.9	1.0	66.7	0.6
Cynarin	0.0	0.0	18.3	0.5
Total dicaffeoylquinic acids	309.7	21.8	396.3	8.4
Luteolin-glycoside	7.5	0.2	7.9	0.3
Apigenin-glycoside	10.0	0.7	9.8	0.1

Table 3. Pharmacokinetic parameters of total chlorogenic acid (CGA), caffeic acid (CA), ferulic acid (FA), dihydrocaffeic acid (DHCA) and dihydroferulic acid (DHFA) absorption in human plasma after consumption of cooked *Cynara scolymus* L. (cultivar Violetto di Provenza) heads

(Mean values and standard deviations)

Parameter	CGA		DHCA		CA		DHFA		FA	
	Mean	SD	Mean	SD	Mean	SD	Mean	SD	Mean	SD
C_{\max} (ng/ml)	6.4	1.7	12.1	7.2	19.5	6.9	21.8	9.2	8.4	4.6
$AUC_{0-\text{last}}$ (ng/ml per h)	9.8	1.6	21.2	6.1	116.1	11.9	58.9	12.3	45.7	5.1
t_{\max} (h)	0.7	0.3	8.0	0.0	1.3	0.7	7.2	1.8	1.1	0.5

 C_{\max} , maximum concentration; $AUC_{0-\text{last}}$, area under the curve from 0 to the last sampling time; t_{\max} , time to reach C_{\max} .

until analysis. In all plasma samples, the following target compounds: CGA, CA, DHCA, FA, DHFA, apigenin and luteolin were analyzed in duplicate with and without enzymatic hydrolysis to detect free and conjugated form. Single phenol levels were evaluated by HPLC with electrochemical detector before and after enzymatic hydrolysis as described by Bugianesi *et al.* (2004) with slight modification. Briefly for free forms, 100 μ l ascorbic acid solution (1%) was added to 0.5 ml plasma and after vortexing and protein precipitation by 500 μ l 3 mol/l HCl–MeOH (1:1, v/v), free phenols were extracted by adding 2 ml ethyl acetate, followed by stirring and sonication (2–3 min) before centrifugation at 1700 g for 5 min. The extraction procedure was repeated twice and the two organic layers were combined and evaporated. The residue was dissolved into 200 μ l mobile phase (phosphate buffer pH 2.8 and methanol (1:1, v/v)). Total phenols were evaluated after adding 100 μ l ascorbic acid solution (1%), 100 μ l 0.2 M acetic acid and 50 μ l β -glucuronidase Type HP2 from Helix Pomatia (Sigma-Aldrich Srl, Milan, Italy). After incubating at 37°C for 1 h proteins were precipitated by adding 700 μ l 3 mol/l HCl–MeOH (1:1, v/v), and total plasma phenols were extracted as previously described for free forms. Blank reagents were performed for free and total phenols forms evaluation. Quantitative analyses were performed on 50 μ l extract using an ESA–HPLC system (ESA, Chelmsford, MA, USA). The HPLC system used consisted of an ESA MODEL 540 refrigerated autoinjector (4°C), an ESA MODEL 580 solvent delivery module with two pumps, an ESA 5600 eight-channels coulometric electrode array detector and the ESA coullarray operating software which controlled all the equipment and carried out data processing. A Supelcosil LC-18 column (25 cm \times 4.6 mm, 5 μ m) with a Perisorb Supelguard LC-18 (Supelco, Milan, Italy) was used. Chromatography was performed at 30°C, at a flow rate of 0.8 ml/min using the following solvent system: solvent A, 0.02 mol NaH₂ PO₄–H₂O adjusted to pH 2.8 with 85% orthophosphoric acid; solvent B, methanol. The linear gradient used consisted of 10% solvent B, increasing to 30% over 7 min and to 33% over 28.5 min, increasing to 45% over 19.5 min, held for 8.5 min and reaching the final condition of 100% 24 min later, after that returning to 10% solvent B over 3 min and maintaining this condition for 4 min. The flow rate of the eluent was constant at 0.8 ml/min and the setting potentials were 60, 120, 200, 340, 480, 620, 760 and 900 mV. Sample peaks were analyzed by matching target peaks with standard peaks on the basis of their retention time and on accuracy

ratio between adjacent channels. The correlation coefficient for all calibration curves was >0.99. Spiked samples were processed and analyzed exactly as described earlier. Mean recovery of polyphenol standards added to plasma (n 3) was >80–118% for CGA, DHCA, CA, DHFA and FA. The method showed a good reproducibility with coefficient of variation within days <5% and between days <9%. Limits of detection were 2.5 ng/ml for CGA, DHCA, DHFA and FA and 2.7 ng/ml for CA.

Statistics

Data are given as the mean and standard deviation. Statistical analysis was performed using the non-parametric Friedman ANOVA test and the Wilcoxon matched pairs test. Differences were considered significant at $P < 0.05$. The computer program used was STATISTICA for Windows (release 4.5; StatSoft Inc., Vigonza PD, Italy).

Results

The composition of both raw and cooked artichoke heads is reported in Table 2. In order to evaluate the effect of domestic cooking on the content of polyphenols, the artichoke heads were analyzed before and after cooking, observing slight increase in CGA content. In addition, after cooking, the increasing of mono- and di-caffeoylquinic acids and the appearance of cynarin were observed. There were no differences in luteolin-glycoside and apigenin-glycoside content. The evaluation of polyphenol content in raw food has been widely studied and optimized on different food matrixes (Hertog *et al.* 1992). Several studies have shown the influence of heating treatments on antioxidant content and their stability (Brenes *et al.* 2002; Lombard *et al.* 2005; Innocenti *et al.* 2005; Rubinskiene *et al.* 2005; Budic-Leto *et al.* 2006).

Fig. 1 shows the plasma concentrations of target compounds after consumption of cooked artichoke heads. Maximum concentration of 6.4 (SD 1.8) ng/ml for CGA was reached after 1 h and declined until its disappearance within 2 h ($P < 0.05$).

Peak plasma concentrations of 19.5 (SD 6.9) ng/ml for total CA were reached within 1 h while FA plasma concentrations showed a biphasic profile with 6.4 (SD 1.5) ng/ml and 8.4 (SD 4.6) ng/ml within 1 h and after 8 h respectively. We observed a significant ($P < 0.05$) increase with respect to baseline for both CA and FA after administration of cooked head of artichoke during the entire experimental time.

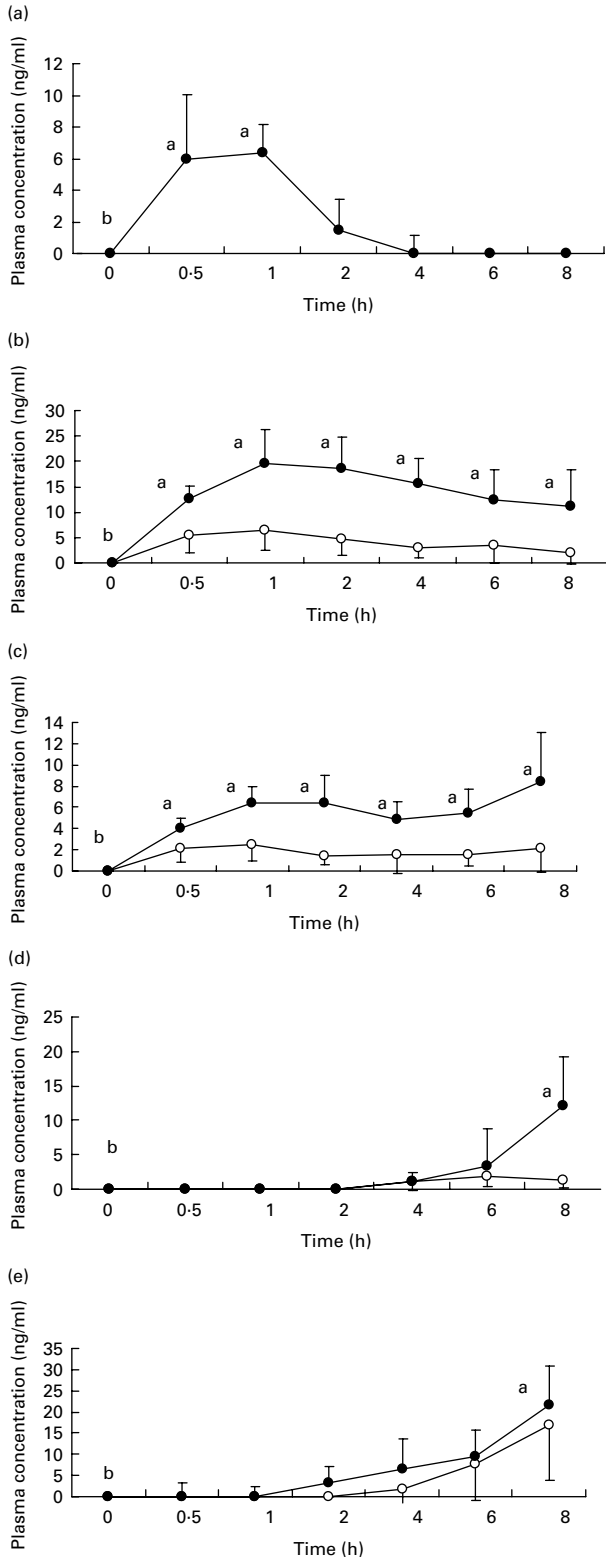


Fig. 1. Plasma concentrations of (a) chlorogenic acid; (b) caffeic acid; (c) ferulic acid; (d) dihydrocaffeic acid and (e) dihydroferulic acid after consumption of cooked *Cyanara scolymus* L. (cultivar Violetta di Provenza) heads. Values are means for five subjects with standard deviations shown by vertical bars. ○, plasma concentration of unconjugated polyphenols; ●, plasma concentration of conjugated plus unconjugated polyphenols. ^{a,b}Wilcoxon test for comparison a v. b, $P < 0.05$.

Increase of DHCA and DHFA total levels was observed within 8 h, to 12.1 (SD 7.2) ng/ml ($P < 0.05$) and 21.8 (SD 9.2) ng/ml ($P < 0.05$) respectively.

Representative chromatograms in Fig. 2 show the absorption profiles in one subject of target artichoke compounds at baseline and during the entire experimental time compared to specific standards. The unnamed peaks have to be considered unknown.

No circulating plasma levels of luteolin and apigenin were present.

Discussion

The current study shows that some artichoke compounds are able to cross the gastric barrier; our findings demonstrate that artichoke target compounds, after crossing the intestinal barrier, reach biologically effective concentrations in the bloodstream. It is well known that most classes of polyphenols are sufficiently absorbed to have the potential to exert biological effects. For example, quercetin after consumption of a meal containing onions, catechins after tea consumption, isoflavones after soya consumption and caffeic acid after drinking red wine, reach micromolar concentrations in blood (Abu Amsha Caccetta *et al.* 1996; Hollman *et al.* 1997; Maiani *et al.* 1997; Scalbert & Williamson, 2000; Manach *et al.* 2002). Bioavailability depends on several endogenous and exogenous factors; the endogenous seem to be related to food matrix, size, chemical structure and ingested amount, the exogenous to digestive enzyme activities and bile excretion, and biotransformations related to liver, kidney, gastrointestinal epithelium and gut microbial flora.

The mean time to reach maximum concentration of 6.4 (SD 1.7) ng/ml for CGA is surprising. Some authors have observed a fast absorption of CGA after oral administration of pure CGA and honeysuckle flower extract in rat (Lafay *et al.* 2006) and rabbit (Yang *et al.* 2004) respectively. Unmodified CGA also appears in human urine (Cremin *et al.* 2001; Olthof *et al.* 2001; Ito *et al.* 2005). On the contrary, other authors have failed to detect intact CGA in plasma after its administration as a pure compound in rats (Azuma *et al.* 2000) or after ingestion of coffee in human subjects (Nardini *et al.* 2002). The rapid plasma detection of CGA could be due to absorption of intact CGA through the stomach or the small intestine barrier. Lafay *et al.* (2006) have identified unchanged CGA in both the gastric vein and aorta showing that the stomach could be involved in CGA absorption in rats. Olthof *et al.* (2001) have indicated that one third of ingested CGA was absorbed in the small intestine in human subjects. Nevertheless, our low plasma CGA levels suggest that the largest portion of CGA and caffeoylquinic acids have undergone biotransformations.

CGA and caffeoylquinic acid hydrolysis, and their metabolites release, could begin in the small intestine. In addition, Andreasen *et al.* (2001) have shown that esterases with activity similar to esters of the major dietary hydroxycinnamates are distributed throughout the intestinal tract of mammals.

The observed CA and FA plasma levels could originate from hydrolysis of both CGA and caffeoylquinic acids, that represent the predominant polyphenol forms in the test meal.

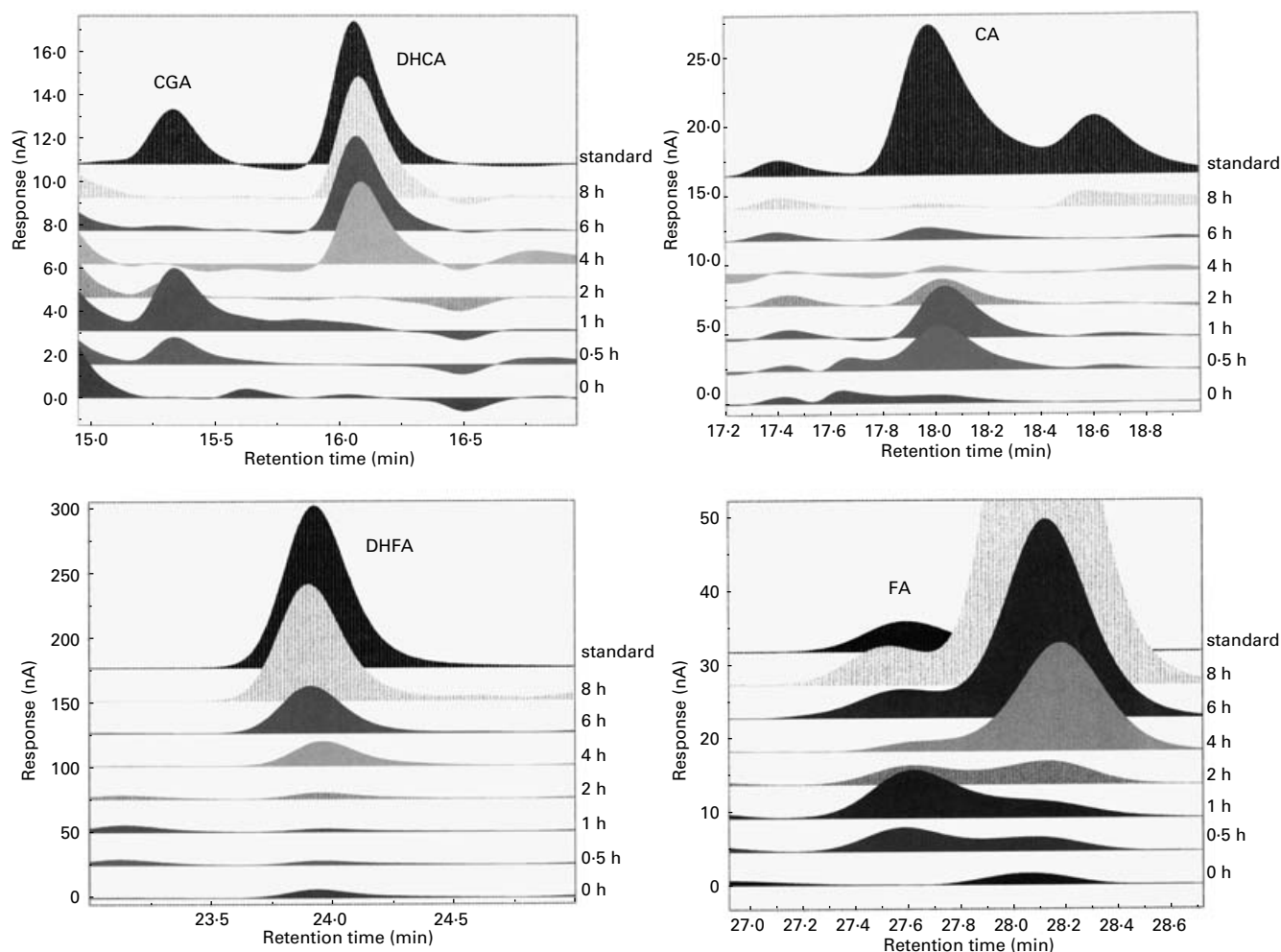


Fig. 2. Representative chromatograms of the absorption profiles in one subject of target artichoke compounds (CGA, chlorogenic acid; DHCA, dihydrocaffeic acid; CA, caffeic acid; DHFA, dihydroferulic acid; FA, ferulic acid) at baseline and during the entire experimental time compared to specific standards. The unnamed peaks have to be considered unknown.

Free plasma levels were low with respect to total forms of CA and FA (Wittemer *et al.* 2005); CA and FA conjugates reached the maximum concentration within 1.5 h. This result seems to suggest that the hydrolysis of active compounds of artichoke heads should have partially occurred in the small intestine (Lafay *et al.* 2006; Azuma *et al.* 2000; Nardini *et al.* 2002).

On reaching the large intestine, unchanged CGA is hydrolyzed by colon enzymes into various aromatic acid metabolites before absorption (*m*-coumaric acid, derivatives of phenylpropionic and benzoic acids; Booth *et al.* 1957; Rechner *et al.* 2001a). Several metabolites have been identified after ingestion of CGA or CA, namely FA, isoFA, DHFA and vanillic acid in human subjects (Rechner *et al.* 2001b).

The colonic microflora has been designated as the major metabolic site for the reactions of the cleavage of ester or glycoside bonds, releasing free hydroxycinnamic acids (Plumb *et al.* 1999; Couteau *et al.* 2001; Rechner *et al.* 2001a).

CA, released by gut microflora, was absorbed and transported to the liver for *o*-methylation to FA (Masri *et al.* 1964; Scheline, 1991; Gumbinger *et al.* 1993; Moridani *et al.* 2001). Nevertheless the gut microflora could be able to

catalyze the *o*-methylation of CA to yield FA and isoFA (Chesson *et al.* 1999).

From our findings, the mean time to reach maximum concentration for DHCA and DHFA was 6–8 h (Wittemer *et al.* 2005). The decrease of CA levels within 6 h and the simultaneous increase of DHCA and DHFA in the bloodstream seem to suggest that CA, released after hydrolysing CGA and caffeoylquinic acids by microflora, was metabolized into DHCA before absorption (Peppercorn & Goldman, 1971). This result could indicate the gut as probable and predominant location of hydroxycinnamate esters metabolism (Andreasen *et al.* 2001, Couteau *et al.* 2001).

Absorbed DHCA could be methylated into DHFA and then dehydrogenated into FA in the liver (Moridani *et al.* 2001). Nevertheless our results indicate that the same reaction mechanisms could take place in the large intestine, particularly dehydrogenation by the colon flora and methylation inside the colon wall. This may contribute to explaining the biphasic concentration profile observed in our samples for total FA.

Moridani *et al.* (2002) have reported that CA and DHCA biotransforming reactions, such as *o*-methylation, GSH conjugation, hydrogenation and dehydrogenation, take places in the

liver catalyzed by P450 cytochrome. On the other hand, some researchers suggest that CA and DHCA methylation into FA and DHFA respectively may occur in the gut (Booth *et al.* 1957; Chesson *et al.* 1999; Ranganathan & Ramasarma, 1974). In both rats and human subjects, dietary plant phenolics can undergo metabolism to form reactive intermediates by catechol-*o*-methyltransferases present in the liver, also in the kidneys and gastrointestinal tracts both stomach and intestine (Nissinen *et al.* 1988; Schultz & Nissinen, 1989; Mannisto *et al.* 1992).

The lack of circulating levels of luteolin and apigenin could be due to their low concentrations in the test meal administered. Several factors may explain the variability of the polyphenols bioavailability, such as the food matrix or background diet. In addition inter-individual variations are also important, because some people might have different levels of metabolizing enzymes or transporters, enabling more efficient absorption of bioactive compounds.

In conclusion, our study shows for the first time the absorption pathways of hydroxycinnamic acids after consumption of edible cooked artichoke in human subjects. These results should be supported by other investigations to confirm the biological activity of cooked artichoke in human subjects and to better understand the mechanism of food phenolic metabolism. With the current conflict existing in the understanding of the metabolism of hydroxycinnamic acids further research is required.

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