In situ **transmission electron energy loss spectroscopy of e-beam modifications in oxide glasses and crystals**

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Electrostatic field-enhanced migration of ions in oxide glasses and crystals has been intensively observed in scanning transmission electron microscopy (STEM) [1]. In this case, excitation by fast electrons causes that some electrons in the atoms are driven out of the trajectory of the beam. The trajectory region thus becomes charged positively and positive ions are repelled from its vicinity [2]. The dynamics of ion migration can be directly observed using *in situ* electron energy loss spectroscopy (EELS) technique, which has been able to reveal a wealth of information about electronic and atomic structure, and chemistry of interaction region [3]. In contrast to its influential role in exploring microstructures of crystalline materials, electron microscopy has little impact on understanding of amorphous materials. Recently, EELS technique has been applied to the study of long-range structural fluctuation in glasses [4]. In this work, we demonstrate that *in situ* EELS technique can also play an important role in the study of glass structures.

Fig. 1 shows the *in situ* EELS of O K-edge in crystalline CaAl₂O₄. The sharp peak at about 528 eV is from O_2 , which is induced during electron irradiation. According to electronic structure calculations (not shown here), the broad peak between 530 and 535eV is mainly due to $Ca - O$ interaction, while feature between 535 and 540 is mainly due to $Al - O$ interaction. The intensity profile of O_2 peak, as well as Ca L_{23} -edge, is also given in Fig. 1. It is noted that the phenomena of ion migrations shown in Fig. 1 are the exactly same with the observations in other glasses (e.g. CaO-Al₂O₃-SiO₂, K₂O-SiO₂ and ZnO-B₂O₃-SiO₂). The details of dynamics have been discussed in [3]. Here two additional features need to be pointed out. One is that nearly half amount of Ca still remains in the illuminated region after rearrangement of O ceases (i.e. the change of O K-edge stops). The question is: why do some Ca ions prefer to be repelled by electrostatic field but others not? The other is that the threshold of O K-edge shifts upwards about 0.5eV. The thresholds of the O K-edge are about 530.8 in the initial spectrum and 531.3eV after 190s of irradiation. The intensity profile of O K-edge in the threshold region (integrated from 530 to 532eV) is compared with the profile of Ca in Fig. 1. Interestingly, the threshold intensity drops significantly within about 70s of irradiation, and becomes relatively constant thereafter. At this point, the irradiated region loses about one third of Ca.

In CaAl₂O₄, two third Ca is 6-fold coordination Ca(6) and one third is 9-fold Ca(9). Correspondingly, O can also be categorized into three groups: $O[Ca(6)]$ is bound to $Ca(6)$ only, O[Ca(9)] is bound to Ca(9) only and the others are bound to both. (Each O atom is also bound to two Al.) Fig. 2 shows the calculations of density of states (DOS) projected on two different types of Ca and O. It is seen that the threshold of the total DOS on Ca(9) is about 0.5eV lower than that on Ca(6). Correspondingly, the threshold of O p-DOS on O[Ca(9)] is also about 0.5eV lower than that on $O[Ca(6)]$. Therefore, we can assign the threshold

intensity of O K-edge (530~532eV) to the oxygen that is bound to 9-fold coordinated Ca. Eliminating this type of oxygen is associated with the repulsion of Ca(9) by electrostatic field. In conclusion, the 9-fold Ca can be repelled more easily than the 6-fold Ca. This criterion can be extended to glasses to identify cation coordination, which is crucial to the measurement of medium-range order in glasses.

References

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Fig. 1 *In situ* EELS of O K-edge using STEM (left) and EELS intensity profiles (right). Energy resolution is about 0.8eV. The arrows in the left indicate the thresholds of the O Kedge and in the right indicate the point that the drop of threshold intensity stops. The dotted lines are guide for eyes.

Fig. 2 Calculations of the total DOS projected on Ca and p-DOS on O in $CaAl₂O₄$ using Wien code. The experimental lattice parameters atom positions are used. Other parameters in the calculations are $K_{\text{max}} \cdot R_{\text{MT}} = 6$ and 56 k-points selected.