

Radial Distribution Function Analyses of Amorphous Carbon Films Containing Silicon and Hydrogen by Energy-Filtered Diffraction and EXELFS

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Short-range order in amorphous materials is most commonly characterized with the use of radial distribution functions (RDFs). Two analytical electron microscopy methods were used in this study to measure RDFs from amorphous carbon films containing different levels of silicon and hydrogen (Si-aC:H): energy-filtered convergent-beam electron diffraction (EFCBED) and extended electron energy-loss fine structure (EXELFS) analyses. The films [1] were deposited in an industrial-scale system onto a thin adhesive titanium layer on silicon substrates by reactive sputtering of carbon with a feed gas of tetramethyl silane (TMS) and argon, to produce a series of films with different Si and H contents (Si/C = 0, 0.04, 0.10, and 0.18).

Cross-section TEM specimens were prepared with a Hitachi FB-2000A focused ion beam (FIB) tool (see Fig. 1). To give enough signal without problems of severe multiple scattering, regions with $t/\lambda \approx 0.3$, where t is specimen thickness and λ is the inelastic scattering mean-free path, were selected for analysis. For specimens with Si/C = 0.04 and 0.10, EFCBED data were acquired with a Philips CM200FEG equipped with a Gatan imaging filter (GIF), operated at nominally 200 kV with the GIF in image mode and a 10-eV slit centered on the zero-loss peak. The selected-area diffraction (SAD) mode was not used because of the well-known aberration-induced shifts of the selected area at large scattering angles. Rather, regions were selected with a probe spread to ~ 600 nm diameter by over-focusing the second condenser lens, resulting in a beam convergence semi-angle $\alpha = 1.3$ mrad. The probe current was ~ 2 nA to limit beam damage (no such effects were observed). Since at the lowest distortion-free camera length the angular range is limited, post-specimen deflection coils were used to precisely shift diffraction patterns for recording with the GIF CCD camera. Intensity profiles were subsequently spliced together to obtain intensity plots for $s = 0$ to 30 nm^{-1} ($s = 2\sin\theta/\lambda$, where now λ is the electron wavelength and 2θ is the scattering angle). The s scale was calibrated from $\langle 011 \rangle$ Si diffraction patterns from the substrate. The data reduction followed that of Cockayne et al. [2] and details can be found elsewhere [3]. Briefly, reduced scattered intensity functions were extracted and Fourier sine transformed to yield reduced density functions $G(r)$, as shown in Fig. 2. With knowledge of the average atomic density ρ_0 from measured plasmon energies [4], the RDF $J(r)$ is obtained, Fig. 3. Electron energy-loss spectra (EELS) for carbon-K EXELFS analysis were acquired in diffraction mode at 300 kV with a Philips CM30 equipped with a LaB_6 cathode and GIF. A 1-nA probe ~ 400 nm in diameter was used with convergence semi-angle $\alpha = 1.6$ mrad, and a collection semi-angle $\beta = 8.2$ mrad. Standard EXELFS data processing steps [5] were followed to extract partial RDFs: (1) Fourier-ratio single scattering deconvolution with Gatan EL/P software; (2) inverse power-law fit to extract EXELFS oscillations from 310 to 525 eV (range limited by 2 to 4 at.% oxygen in specimens); (3) oscillation function $\chi(k)$ for k from 27 to 80 nm^{-1} generated with the energy threshold held constant at 283 eV; and (4) Fast Fourier Transform (FFT) of $k^2\chi(k)$ to yield a partial RDF with a phase shift $\phi = 0.038$ nm applied (based on Teo and Lee [6] method), as shown in Fig. 4.

Inspection, analysis, and interpretation of the EFCBED results (Figs. 2 & 3) lead to the following conclusions [4]. Peaks in $G(r)$ for Si/C = 0.04 are narrower with a higher peak-to-width ratio than those for Si/C = 0.10. The near-neighbor distances suggest the presence of amorphous hydrogenated SiC domains. Peaks in $J(r)$ for Si/C = 0.04 are at 0.16 and 0.28 nm, suggesting the most prominent average bond angle is slightly higher than that for sp^2 trigonally bonded C (122° versus 120°) and consistent with a slightly distorted graphitic ring structure. The second shell peak in $J(r)$ for Si/C =

0.10 has a plateau-like shape, indicative of a variety of discrete bond lengths; the 0.19-0.31 nm pair indicates an average bond angle of 109° , possibly consistent with the presence of C-Si tetrahedral bonding, as in crystalline SiC. The ratio of coordination numbers for the second and first atomic shells obtained from $J(r)$ is 2.7 for both Si/C = 0.04 and 0.10, which is intermediate to values for diamond (3.0) and graphite (2.0). The EXELFS results (Fig. 4) lead to the following conclusions [4]. The first nearest-neighbor distance for Si/C = 0.04 is ~ 0.015 nm less than that in the other films and is attributed to compressive stress effects. The ratio of second to first shell intensities is higher for Si/C = 0.04 than for the other films, suggesting highly interlinked C and Si around C atoms. Increased deviations from the average bond length are indicated by the wider first nearest-neighbor peak for Si/C = 0.0 and 0.04. The second nearest-neighbor shells for Si/C = 0.0 and 0.04 extend past the 0.246, 0.285, and 0.335 nm positions of crystalline graphite, possibly suggesting any graphitic content is distorted. The second nearest-neighbor shells for Si/C = 0.10 and 0.18 align well with peak positions expected for graphite, suggesting that less-distorted graphite rings are present [7].

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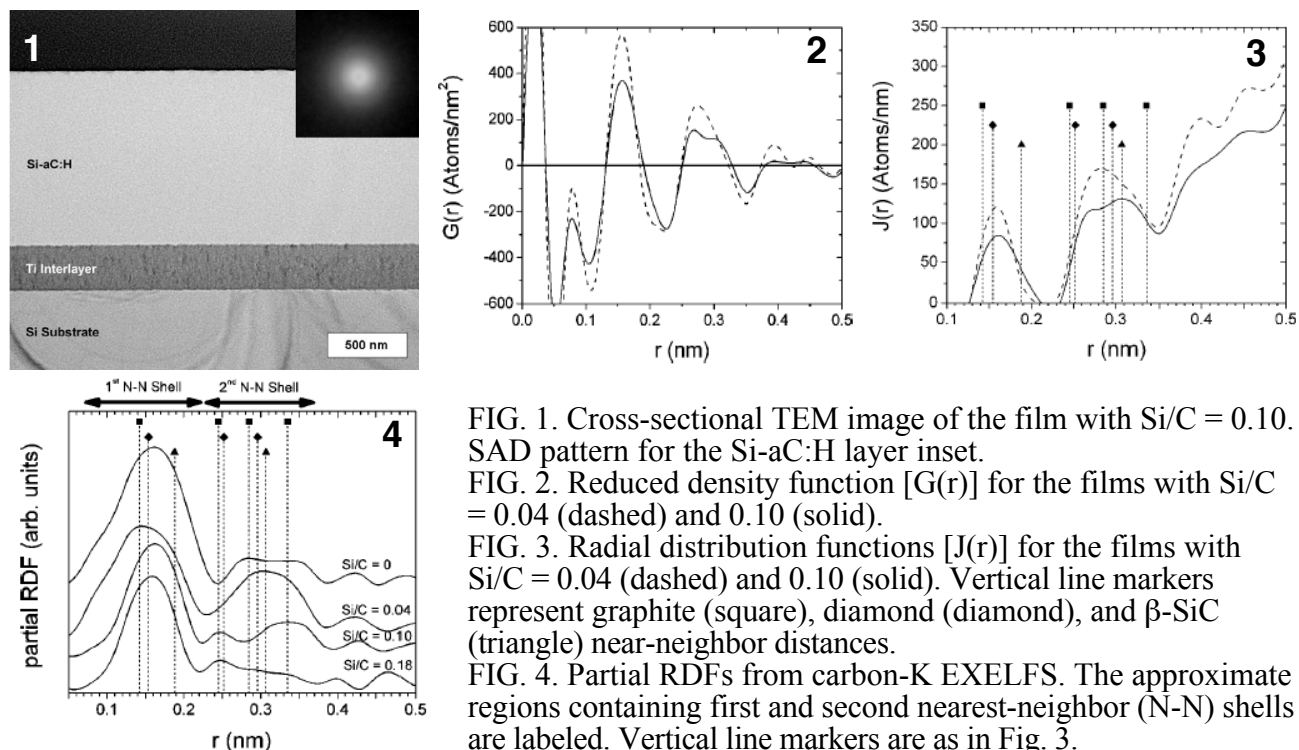


FIG. 1. Cross-sectional TEM image of the film with Si/C = 0.10. SAD pattern for the Si-aC:H layer inset.

FIG. 2. Reduced density function [$G(r)$] for the films with Si/C = 0.04 (dashed) and 0.10 (solid).

FIG. 3. Radial distribution functions [$J(r)$] for the films with Si/C = 0.04 (dashed) and 0.10 (solid). Vertical line markers represent graphite (square), diamond (diamond), and β -SiC (triangle) near-neighbor distances.

FIG. 4. Partial RDFs from carbon-K EXELFS. The approximate regions containing first and second nearest-neighbor (N-N) shells are labeled. Vertical line markers are as in Fig. 3.