

Direct Observation of Atomic Surface Structures of CeO₂ Nanoparticles

Jianguo Wen¹, Yuyuan Lin², Zili Wu³, Kenneth R. Poeppelmeier^{2,4}, and Laurence D. Marks²

¹Electron Microscopy Center, Argonne National Laboratory, Argonne, IL 60439

²Department of Materials Science and Engineering, Northwestern University, Evanston, IL 60208

³Chemical Science Division & Center for Nanophase Materials Sciences, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 37831

⁴Department of Chemistry, Northwestern University, Evanston, IL 60208

CeO₂ has a wide range of applications such as catalysts and as electrodes in solid oxide fuel cells due to its unique physical, chemical and electrochemical properties [1,2]. Many of these applications are related to surface structures of CeO₂. For example CeO₂ nanoparticles are used as a catalytic support since surface oxygen vacancies are expected to promote catalytic metal particle dispersion [1]. However, due to the lack of clear observation of oxygen columns using TEM or scanning TEM, atomic surface structures of CeO₂ nanoparticles are under debate. Using high resolution electron microscopy with Argonne Chromatic Aberration-corrected TEM, we are able to observe both Ce and O columns, allowing us to directly observe atomic structures of the (100), (110) and (111) surfaces of CeO₂ nanoparticles.

CeO₂ nanocubes consists 6 dominant flat (100) surfaces, 12 edge (110) surfaces, and 8 corner (111) surfaces. Fig. 1 shows HREM images of the predominantly exposed (100) surface along [110] zone axis. Fig. 1b shows two distinctive surface structures on (100) surface, one is a surface with a Ce termination (yellow bar) and another surface with an O termination (red bar). Thus both O and Ce terminations can exist on the same surface. (100) surface typically has a mixture of Ce, O, and reduced CeO terminations (Fig. 1e). HREM simulation indicates that partial occupancy of the outermost layer occurs at both Ce and O positions, resulting the exposure of the subsurface layer. During HREM observation, the hopping of atoms on the surface is often observed even when the electron beam intensity is reduced to $5 \times 10^2 \text{ e}/\text{\AA}^2\text{s}$ [3].

HREM image in Fig. 2a shows that the (110) surface consists of flat CeO_{2-x} terminations and small amount of “saw-like” (111) nanofacets indicated by white arrows. For the flat part of the (110) surface, partial surface oxygen vacancies result in two types of termination layers: (110) surface with a CeO₂ surface termination and (110) surface with a Ce termination as supported in simulated images in Fig. 2b and 2c. A line profile of the surface layer from A1 to A2 shows more clearly the existence of many surface O vacancies. We observed an average of approximately 30% oxygen vacancies on the flat (110) surface of CeO₂ nanocubes.

The (111) facet is determined as O-terminated by comparing simulated HREM image with the O-terminated surface with the experimental surface contrast, which is in agreement with the previous IR studies and electrostatic considerations for the surface HREM observations of all three surfaces indicate that surface reconstruction, nanofacet, and surface vacancies need to be considered when establishing the structure-catalysis relationship for CeO₂ nanoparticles [4].

References:

- [1] Trovarelli, A. “Catalytic properties of ceria and CeO₂-containing materials”, *Catalysis Reviews* 38, 439 (1996).
- [2] Mogensen, M.; Sammes, N. M.; Tompsett, G. A. “Physical, chemical and electrochemical properties of pure and doped ceria”, *Solid State Ionics* 29, (1–4), 63-94, (2000).

- [3] Lin, Y., Wu, Z., Wen, J., Poeppelmeier, K.R., Marks, L.D., "Imaging the atomic surface structures of CeO₂ nanoparticles", *Nano Lett* 14 (1), 191-196 (2013).
- [4] We acknowledge funding from Northwestern University Institute for Catalysis in Energy Processes (ICEP) on grant number DOE DE-FG02-03-ER15457 (Y.L., K.R.P. and L.D.M.). The electron microscopy was performed at the Electron Microscopy Center at Argonne National Laboratory, a U.S. Department of Energy Office of Science Laboratory operated under Contract No. DE-AC02-06CH11357 by UChicago Argonne, LLC. The synthesis of ceria nanocubes was conducted at Oak Ridge National Laboratory and was sponsored by the Division of Chemical Sciences, Geosciences, and Biosciences, Office of Basic Energy Sciences, U.S. Department of Energy.

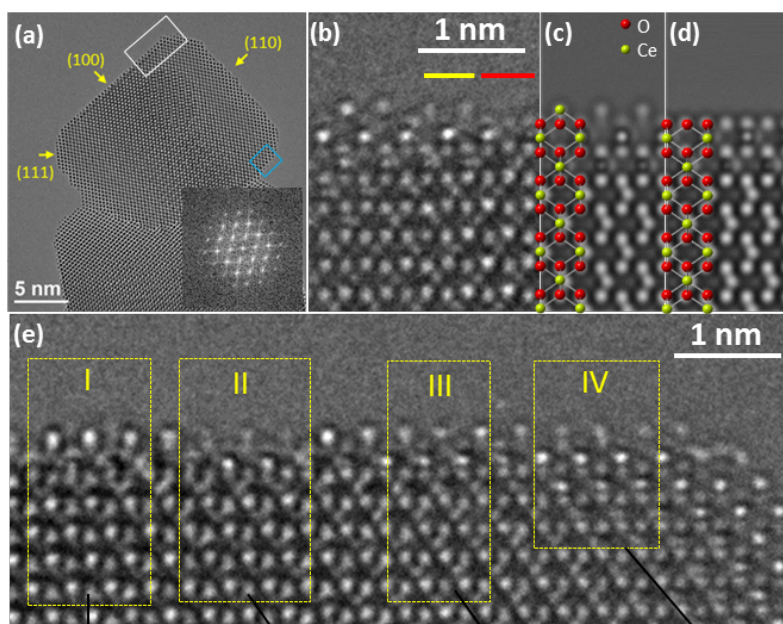


Fig. 1. Atomic structure of (100) surfaces of CeO₂ nanocubes. (a) A HREM image of a typical CeO₂ nanocube along [110]. (b) A magnified HREM image of the (100) surface of (a), highlighted with the blue box. (c)-(d) Simulated HREM images of (b) using Ce and O terminations respectively. (e) A magnified HREM image of the (100) surface of (a), highlighted with the white box. Regions I-IV shows the (100) surfaces with different terminations or occupancies.

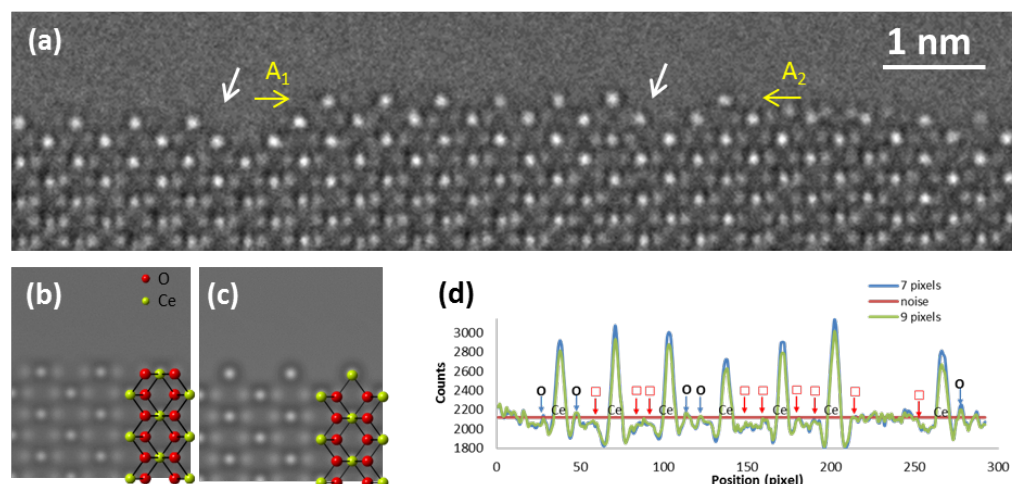


Fig. 2. (110) surfaces of CeO₂ nanocubes. (a) An experimental HREM image on a (110) surface of CeO₂ nanocubes along [110]. The white arrows indicate (111) nanofacets. (b) A simulated HREM image of the CeO₂ (110) surface with a CeO₂ surface termination. (c) A simulated HREM image of the CeO₂ (110) surface with a Ce termination. (d) Integrated line profiles from A1 to A2 indicated in (a). O vacancies are indicated by the squares (□).