

Atomic Surface Structures of Oxide Nanoparticles with Well-defined Shapes

Jianguo Wen¹, Yuyuan Lin², Huaping Sheng¹, Lifen Wang¹, Dean J. Miller¹, Zili Wu³, Kenneth R. Poeppelmeier^{2,4}, and Laurence D. Marks²

¹Center for Nanoscale Materials, Argonne National Laboratory, Argonne, IL

²Department of Materials Science & Engineering, Northwestern University, Evanston, IL

³Chemical Science Division & Center for Nanophase Materials Sciences, Oak Ridge National Laboratory, Oak Ridge, Tennessee,

⁴Department of Chemistry, Northwestern University, Evanston, IL

Recent studies have shown that catalytic activities can be tuned by controlling the shape of nanoparticles such as SrTiO₃, CeO₂, and Co₃O₄ [1]. Therefore, determination of surface structure is very important to understand structure-property relationships for these oxide nanoparticles. The Argonne Chromatic-corrected TEM (ACAT) has an image corrector that corrects both spherical (C_s) and chromatic aberration (C_c). C_c correction allows the correction of C_s towards zero to improve resolution without compromising contrast. Using this unique feature, we correct both C_s and C_c to small values to achieve direct structure interpretable HREM images including oxygen atomic columns. In this study, atomic surface structures of SrTiO₃, CeO₂, Co₃O₄ nanocubes are observed by using aberration-corrected HREM.

As-prepared SrTiO₃ nanocubes have 6 well-defined {100} surfaces. We have shown previously that by tilting SrTiO₃ nanocubes to <110> directions, with the oxygen atoms clearly observed using aberration-corrected HREM, we are able to determine surface atomic structure on (100) surface of SrTiO₃ nanocubes. HRTEM studies show that the (100) surface of SrTiO₃ nanocubes can be SrO, TiO₂-rich reconstructions, or mixed with SrO and TiO₂-rich reconstructions depending on synthetic procedures [2].

CeO₂ nanocubes consists of 6 dominant flat {100} surfaces, 12 edge {110} surfaces, and 8 corner {111} surfaces. Viewing along <110> zone axis allows one to observe atomic structure on (100), (001) and (111) surfaces simultaneously [3]. The (100) surface has a mixture of Ce, O and reduced CeO terminations on the outermost surface. The (110) surface has a combination of reduced flat CeO_{2-x} surface layers and "sawtooth-like" (111) nanofacets. The CeO₂ (111) surface is O-terminated. During HREM observation, the hopping of atoms on the surface is often observed even when the electron beam intensity is reduced to 5×10² e/Å²s. Fig. 1 shows the first layer of (111) surface diffuses away under an electron beam irradiation. Several new surface configurations are observed when Ce and O atoms hop on (100), (100) and (111) surfaces.

Co₃O₄ nanocubes consist of 6 dominant flat {100} surfaces and 8 corner {111} surfaces. The inset in Fig. 2 shows the project of Co₃O₄ along [1-10] direction. Oxygen is indicated by red dots. Tetrahedral and octahedral Co sites are indicated by green and blue dots respectively. HRTEM images in Fig. 2 show that (001) terminates at O-Co (at octahedral sites) and (111) terminates at O-Co (at tetrahedral sites).

SrTiO₃ nanocubes have ideal surfaces free of surface steps and islands. But CeO₂ and Co₃O₄ nanocubes often show complex surface structures such as sawtooth-like structures. We interpret

this because of difference in surface chemistry since the termination layers of SrO and TiO₂ are charge neutral, but it is not the case for CeO₂ and Co₃O₄ nanocubes [4].

References:

- [1] Lin, Y. *et al.*, Nano Lett. **15**, 5375 (2015).
 [2] Lin, Y. *et al.*, Phys. Rev. Lett. **111**, 156101 (2013).
 [3] Lin, Y. *et al.* Nano Lett. **14**, 191 (2013).
 [4] Electron microscopy was performed at the Center for Nanoscale Materials at Argonne National Laboratory, a U.S. Department of Energy Office of Science Laboratory operated under Contract No. DE-AC02-06CH11357 by UChicago Argonne, LLC. Y.L., K.R.P. and L.D.M. acknowledge funding from Northwestern University Institute for Catalysis in Energy Processes (ICEP) on grant number DOE DE-FG02-03-ER15457. The synthesis of ceria nanocubes was conducted at Oak Ridge National Laboratory and was sponsored by the Division of Chemical Sciences, Geosciences, and Biosciences, Office of Basic Energy Sciences, U.S. Department of Energy.

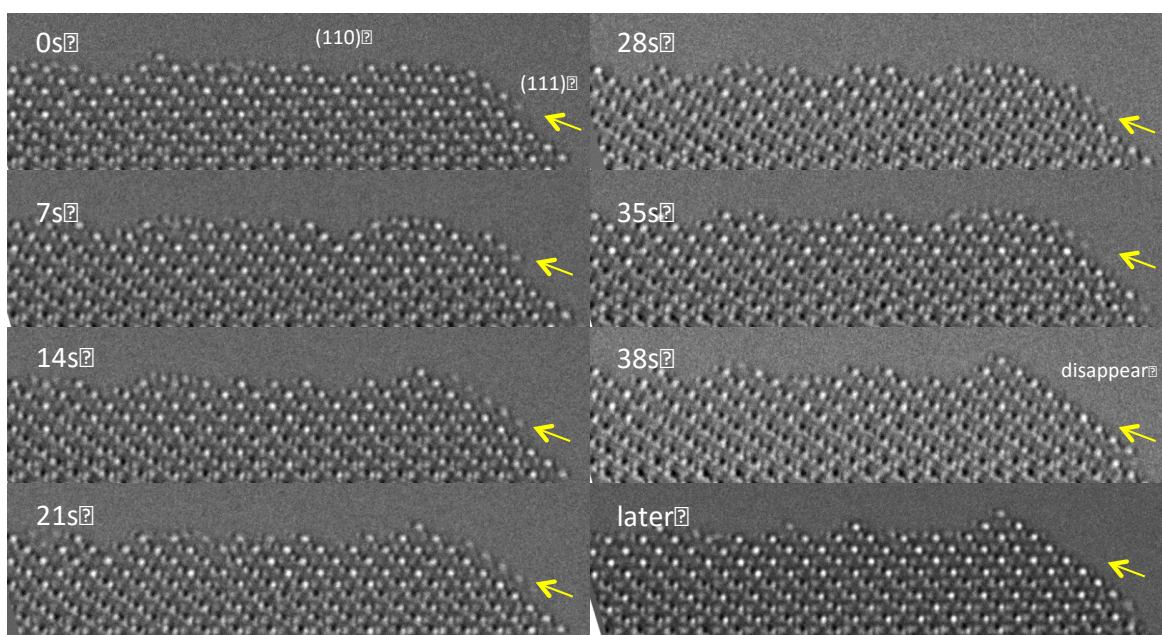


Figure 1. HREM images of $\{110\}/\{111\}$ corner of a CeO₂ nanoparticle in selected time frames. The first layer of CeO₂ (111) surface diffuses away completely under an electron beam irradiation.

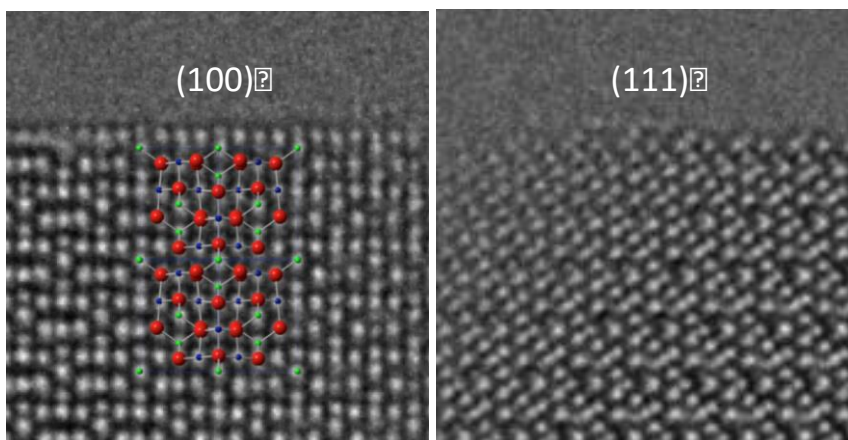


Figure 2. HREM images showing surface atomic structures of Co₃O₄ nanocubes on a) (100), b) (111) surfaces. The red dots represent the oxygen atoms. The blue and green dots represent cobalt at octahedral and tetrahedral sites by atoms with valence +3 and +2.