

FIRST REPORTED SAMPLES FROM THE RADIOCARBON LABORATORY OF THE UNIVERSITY OF TENNESSEE CENTER FOR ARCHAEOMETRY AND GEOCHRONOLOGY: DATES FROM THE MCCROSKY ISLAND ARCHAEOLOGICAL SITE (40SV43), SEVIER COUNTY, TENNESSEE, USA

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ABSTRACT. This study presents the results of archaeological samples submitted for dating at the recently constructed University of Tennessee Center for Archaeometry and Geochronology (UTCAG) radiocarbon dating laboratory (Knoxville, Tennessee, USA). The samples selected for this initial study were obtained from excavations at the McCrosky Island site (40SV43) in Sevier County, Tennessee, USA. Three of the samples dated were split between the UTCAG laboratory and another laboratory to assess the UTCAG laboratory protocols. In an effort to further validate the laboratory methods employed, several other samples were submitted without prior knowledge of contextual data. The dates obtained for these samples were then compared to their association with recovered artifacts and/or archaeological context.

INTRODUCTION

The University of Tennessee Center for Archaeometry and Geochronology (UTCAG) was established in 2001 with initial funding from the University of Tennessee President's Research Initiatives granting program. This award provided the necessary funding to implement the construction of a conventional radiocarbon dating laboratory through a partnership with researchers from the Illinois State Geological Survey (ISGS). Construction of the benzene extraction system, the creation and implementation of operation protocols for the laboratory, and a period of interlaboratory testing were completed in 2005.

The UTCAG laboratory currently utilizes the liquid scintillation counting (LSC) method for dating. Our laboratory is equipped with a benzene synthesis system and a Quantulus 1220™ ultra low-level scintillation counter. We are capable of processing a variety of samples, including charcoal, wood, bone, paleosol, coral, and shell. This facility serves as an additional resource for archaeologists and earth scientists throughout the United States.

In an effort to evaluate the laboratory's performance, a series of interlaboratory comparative samples were analyzed. These samples consisted of remaining portions of the Fourth International Radiocarbon Intercomparison (FIRI) data set, dendrochronologically dated wood samples, and intercomparisons of split samples dated at both ISGS and UTCAG. Once the results from these samples were deemed satisfactory, a blind test of archaeologically obtained samples was initiated to further examine the laboratory's performance. The results of this blind test are reported here.

A series of 15 dates were calculated from 14 archaeological samples obtained from eastern Tennessee, USA. Some of these samples consist of split samples sent to Beta Analytic, Inc. (Miami, Florida, USA). Other samples were obtained from features containing lithic and ceramic artifacts representative of known temporal contexts. These dates are presented as documentation of the efforts at the UTCAG dating laboratory, though periodic evaluations of accuracy are an ongoing component of the laboratory protocol.

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METHODS

Samples

The samples selected for this study were obtained from archaeological investigations of the McCrosky Island site (40SV43), Sevier County, Tennessee (35°55'48"N, 83°36'1"W) (Figure 1). Artifacts recovered from the site indicate that it was occupied from the Early Woodland to Late Mississippian periods, about 800 BC–AD 1600. Sample collection began in 1995 and currently continues (Polhemus 1996). A list of the sample types and year collected is presented in Table 1. Samples were dried before being sent for radiometric analysis at UTCAG or Beta Analytic, Inc.

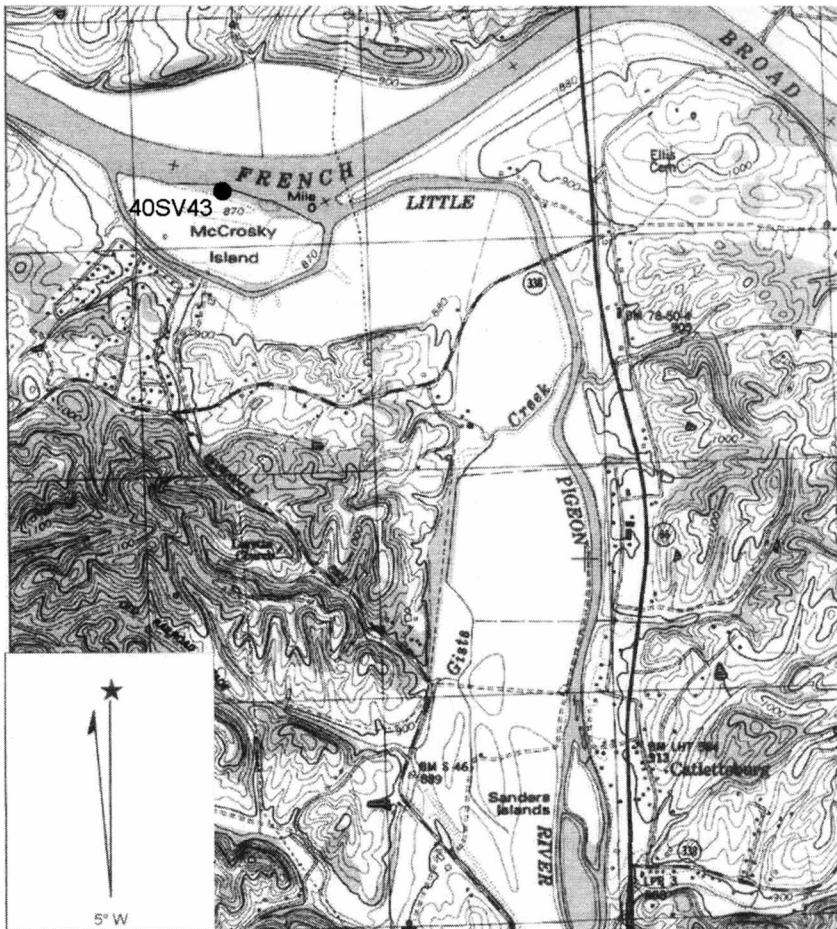


Figure 1 Location of the McCrosky Island archaeological site (40SV43); 1 inch equals 2000 feet.

Sample Pretreatment

Samples were rinsed with deionized water (DI H₂O) and were then inspected under an optical microscope for obvious contaminants such as grass, rootlets, and any other foreign materials. Samples were subsequently dried overnight in an 80 °C oven. Chemical pretreatment of 4 of the samples dated consisted of the acid/alkali/acid (AAA) method (Table 1). Samples were boiled in 2N hydrochloric acid (HCl) for 1 hr in order to remove inorganic carbon contaminants. Upon cooling and set-

Table 1 Samples analyzed by UTCAG.

Sample identification	Sample description	Year collected	UTCAG pretreatment
40SV43C14-5	Charcoal associated with pit feature ^a	2005	Acid only
40SV43C14-6	Charcoal from basin feature	2000	AAA
40SV43C14-7	River cane	1995	AAA
40SV43C14-8	Log associated with feature	1995	AAA
40SV43C14-9	Log associated with pit feature	1995	AAA
40SV43C14-13	Charred tree near riverbank ^a	2005	Acid only
40SV43C14-17	Charcoal associated with midden	2006	Acid only
40SV43C14-18	Charcoal associated with pit feature	1995	Acid only
40SV43C14-19	Charcoal associated with pit feature	1995	Acid only
40SV43C14-20	Charcoal from basin feature	1995	Acid only
40SV43C14-22	Charcoal from basin feature	1995	Acid only
40SV43C14-23	Charcoal from basin feature ^a	2006	Acid only
40SV43C14-24	Charcoal from basin feature	2006	Acid only
40SV43C14-25	Charcoal from pit feature	2006	Acid only

^aIndicates those samples split between UTCAG and Beta Analytic, Inc.

ting of the material, the samples were neutralized by rinsing with DI H₂O. Humic contaminants were removed by boiling the samples in 0.5N sodium hydroxide (NaOH) for 30 min. Once cooled, the samples were rinsed several times with DI H₂O. The samples were then boiled twice with fresh DI H₂O to expel final traces of NaOH absorbed by the material. After boiling, the samples were allowed to cool and again rinsed with DI H₂O. Samples were treated with 5% phosphoric acid (H₃PO₄) for 20 min and subsequently rinsed with DI H₂O to neutrality. Finally, the samples were dried overnight in an 80 °C oven.

The remaining 10 charcoal samples submitted for dating were found to be too friable for the standard AAA pretreatment process (see Table 1). Therefore, only the initial acid pretreatment was performed for these samples, followed by several DI H₂O rinses to return the samples to neutrality. The samples were then dried overnight in an 80 °C oven.

Benzene Synthesis

Benzene synthesis involved 4 primary steps: 1) combustion; 2) lithium carbide formation; 3) acetylene generation; and 4) benzene trimerization. The procedures are based on those established and utilized by the ISGS laboratory (see Noakes et al. 1965, 1967; Coleman et al. 1972). Combustion consisted of a 3-part process. First, the material was combusted in an argon-rich atmosphere at 550 °C. This was followed by combustion in an argon-rich atmosphere at 750 °C. Finally, samples were completely pyrolyzed in an oxygen-rich atmosphere at 750 °C. Benzene synthesis of the oxalic acid II standard was performed under these same conditions. Combustion temperature, however, was maintained at 450 °C for oxalic standard combustion. Throughout this process, gases were filtered through silver nitrate (AgNO₃) and chromic acid (Na₂Cr₂O₇) solutions for removal of residual chlorine (from HCl pretreatment) and sulfur dioxide (SO₂) (generated by oxidation of organic sulfur in samples during combustion), respectively. Excess water was removed by freezing in a trap cooled by a dry ice and isopropyl solution. The carbon dioxide (CO₂) obtained was collected in 3 liquid nitrogen traps. After reevacuation of the system, the CO₂ was sublimated into precalibrated storage ball(s) and a measurement of the CO₂ collected was obtained.

To form lithium carbide (Li_2C_2), lithium metal (2.5 g of lithium for each liter of sample CO_2) was melted by heating to 615 °C in a vacuum-evacuated reactor vessel. Once all of the lithium metal had melted and the excess gases evacuated, sample CO_2 was introduced slowly, making sure that the reactor pressure did not exceed $-20''$ Hg. CO_2 flow continued until the pressure in the reactor and storage ball(s) decreased to $-30''$ Hg and the reactor temperature returned to 615 °C. The reactor temperature was then increased to 850 °C in 30 min and maintained for an additional 45 min to convert any remaining carbon black to lithium carbide (Li_2C_2). The reaction vessel was then evacuated until cooled to remove any remaining unreacted gases.

Acetylene was generated by introducing DI H_2O to the cooled lithium carbide. Excess water was removed from the reaction gases with the aid of 2 traps, cooled with dry ice and isopropyl solutions. The acetylene was collected in 3 liquid nitrogen traps. All other non-condensable gases were evacuated from the system. Acetylene was then sublimated, passed through an additional dry ice/isopropyl trap to remove any remaining H_2O , and collected in the precalibrated storage ball(s).

Trimerization of benzene was achieved by introducing acetylene to a commercially available vanadium oxide catalyst (Coleman et al. 1972). A column containing preconditioned catalyst was activated under vacuum at 350 °C for 2 hr. The column was then allowed to cool to room temperature before the acetylene was introduced. The column was kept cool with an ice-water bath. Once trimerization was completed, the benzene was extracted by heating the column in a tube furnace at 120 °C for 2 hr. The liquid benzene was frozen in a collection vial, cooled with a dry ice/isopropyl solution. The sublimated benzene was then prepared for the liquid scintillation counter.

All samples were counted in specially designed copper and Teflon® vials for use with the Quantulus 1220. A total sample volume of 5 mL was placed in the vial for counting. This consisted of 4 mL of sample, plus 1 mL of scintillation cocktail. In cases where sample volume was less than 4 mL, an appropriate amount of dead benzene was added to achieve the 4 mL sample volume. Each sample was analyzed for 2700 min, consisting of 27 cycles at 100 min each, in a prestandardized counting vial. Each sample was followed by a similar analysis of a background sample, counted in the same prestandardized counting vial.

Analysis of $^{13}\text{C}/^{12}\text{C}$ Ratio

The stable carbon isotopic composition of CO_2 gas analyses were performed by the stable isotope laboratory of the Department of Earth and Planetary Science, University of Tennessee. Samples were analyzed using a Finnigan Delta^{plus} spectrometer. Results are reported relative to the Vienna Pee-Dee belemnite (VPDB) standard with a precision of $\pm 0.02\%$.

RESULTS AND DISCUSSION

Comparison with Beta Analytic, Inc.

Three samples were split for analysis at UTCAG and Beta Analytic, Inc. (40SV43C14-5, 40SV43C14-13, 40SV43C14-23). The results from both laboratories are presented in Table 2 and indicate good agreement between splits of all 3 dated samples. The differences between samples 40SV43C14-5 and 40SV43C14-13 are within 1σ (50 ^{14}C yr and 55 ^{14}C yr, respectively), whereas sample 40SV43C14-23 is within 2σ (130 ^{14}C yr). The dates for sample 40SV43C14-23 provide a chronological marker for Swift Creek Complicated Stamped ceramics (Jennings and Fairbanks 1939; Sears 1952) recovered from this archaeological context.

Table 2 Results of interlaboratory comparison.

Sample identification	UTCAG		Beta Analytic	
	Conventional age (BP ±1 σ)	δ ¹³ C (‰)	Conventional age (BP ±1 σ)	δ ¹³ C (‰)
40SV43C14-5	650 ± 70	-25.27	600 ± 40	-25.4
40SV43C14-13	285 ± 70	-26.75	230 ± 40	-25.6
40SV43C14-23	1640 ± 70	-25.70	1510 ± 40	-25.8

An additional sample, 40SV23C14-17, was not submitted as a split sample, but was associated with an archaeologically contemporaneous midden feature sampled for dating by Beta Analytic (40SV23C14-15). The midden contained early Middle Woodland Connestee phase artifacts (see Ferguson 1971). The UTCAG sample dated to 1840 ± 70 BP, and the Beta sample dated to 1760 ± 40 BP. Although the difference between the conventional dates is 80 ¹⁴C yr and falls within the 2-σ range, the intercept of the 2-σ calibrated ages differs by only 10 yr (cal AD 200 and cal AD 190, respectively). Therefore, it is concluded that the dates support the archaeologically identified contemporaneity. The dates are consistent with the Connestee phase in this region (Ferguson 1971).

Contextual Analysis of Remaining Samples

The reliability of the remaining dates was evaluated by comparison to their respective archaeological contexts, based on stratigraphic position or association with temporally known artifact types. The approximate temporal order of the remaining samples is presented in Table 3. The dates calculated by the UTCAG lab and a brief description of the samples are presented below.

Table 3 Approximate temporal order of remaining UTCAG samples.

Time period	Sample identification	UTCAG #	Sample description
Early Dallas phase	40SV43C14-18	07-005	Charcoal associated with pit feature
	40SV43C14-9	06-024	Log associated with pit feature
	40SV43C14-7	06-021	River cane
	40SV43C14-8	06-019	Log associated with feature
	40SV43C14-20	07-007	Charcoal from basin feature
Emergent Mississippian	40SV43C14-19	07-006	Charcoal associated with pit feature
Late Middle Woodland	40SV43C14-6	06-020	Charcoal from basin feature
Early Middle Woodland	40SV43C14-24	07-002	Charcoal from basin feature
	40SV43C14-25	07-003	Charcoal from pit feature
Early Woodland	40SV43C14-22	07-009	Charcoal from basin feature

UTCAG06-020. McCrosky Is., Fea. 201, 40SV43C14-6 **1400 ± 70**
δ¹³C = -26.3‰

Charcoal from basin containing fire-cracked rock. Basin contains remains of late Middle Woodland period ceramics. Collected 2000 and submitted 2006 by R R Polhemus. This date is consistent with 2 other, stratigraphically contemporaneous features dated by Beta Analytic (Beta #216648 and #216649; 1300 ± 40 BP and 1280 ± 40 BP, respectively).

Comment: This sample provides a date for complicated stamped ceramics from this site that resemble Napier Complicated Stamped ceramics (Jennings and Fairbanks 1939; Sears 1952) but are not stylistically identical.

UTCAG06-021. McCrosky Is., Fea. 40, 40SV43C14-7 **875 ± 70**
 $\delta^{13}\text{C} = -27.2\text{‰}$

Carbonized river cane, associated with Early Dallas phase, Early Mississippian period burial feature. Collected 1995 and submitted 2006 by R R Polhemus. 40SV43C14-8 (UTCAG06-019) is associated with the same feature.

Comment: The date of this sample overlaps UTCAG06-019 at the 1- σ level.

UTCAG06-019. McCrosky Is., Fea. 40, 40SV43C14-8 **770 ± 70**
 $\delta^{13}\text{C} = -27.9\text{‰}$

Carbonized wood, associated with Early Dallas phase, Early Mississippian period burial feature. Collected 1995 and submitted 2006 by R R Polhemus. 40SV43C14-7 (UTCAG06-021) is associated with the same feature.

Comment: The date of this sample overlaps UTCAG06-021 at the 1- σ level.

UTCAG06-024. McCrosky Is., Fea. 158, 40SV43C14-9 **815 ± 70**
 $\delta^{13}\text{C} = -25.6\text{‰}$

Carbonized wood, associated with Early Dallas phase, Early Mississippian period rectangular pit feature. Collected 1995 and submitted 2006 by R R Polhemus. This feature is stratigraphically and artifactually related to features 40, 76, and 236.

UTCAG07-005. McCrosky Is., Fea. 76, 40SV43C14-18 **730 ± 70**
 $\delta^{13}\text{C} = -24.8\text{‰}$

Wood charcoal associated with Early Dallas phase, Early Mississippian period rectangular pit feature. Collected 1995 and submitted 2007 by R R Polhemus. This feature is stratigraphically and artifactually related to features 40, 158, and 236.

UTCAG07-005V1. McCrosky Is., Fea. 76, 40SV43C14-18(2) **790 ± 70**
 $\delta^{13}\text{C} = -24.8\text{‰}$

This sample represents a second aliquot of the benzene extracted from UTCAG07-005. The sample was counted in a separately standardized sample vial and served as a test of the laboratory's vial standardization protocols. The difference in dates is 60 ^{14}C yr, within the 1- σ range.

UTCAG07-006. McCrosky Is., Fea. 120, 40SV43C14-19 **1155 ± 70**
 $\delta^{13}\text{C} = -27.8\text{‰}$

Mixed charcoal sample recovered from ash lens from cylindrical pit feature. Collected 1995 and submitted 2007 by R R Polhemus.

Comment: The date of this sample indicates an emergent Mississippian age for this feature and is consistent with the ceramic artifacts recovered from the feature. Feature 120 contained Mississippian vessel forms constructed of both shell and limestone tempered clays.

UTCAG07-007. McCrosky Is., Fea. 156, 40SV43C14-20 **915 ± 70**
 $\delta^{13}\text{C} = -26.5\text{‰}$

Mixed charcoal sample recovered from fire-cracked rock basin containing charred wood and maize remains. Collected 1995 and submitted 2007 by R R Polhemus. Maize was a dominant crop of Mississippian period peoples.

Comment: This date places this pit feature in the Early Dallas phase, Early Mississippian period.

UTCAG07-009. McCrosky Is., Fea. 149, 40SV43C14-22 2175 ± 70
 $\delta^{13}\text{C} = -26.9\text{‰}$

Mixed charcoal sample recovered from shallow circular basin feature. Collected 1995 and submitted 2007 by R R Polhemus. The feature contained artifacts and stone tools indicative of Early Woodland peoples of the Southeast USA. The date is consistent with these artifact types.

UTCAG07-002. McCrosky Is., Fea. 284, 40SV43C14-24 1680 ± 70
 $\delta^{13}\text{C} = -27.6\text{‰}$

Mixed charcoal sample recovered from fire-cracked rock-filled basin. Collected 2006 and submitted 2007 by R R Polhemus. This feature is from the same context as the features that provided samples 40SV43C14-17 and 40SV43C14-25.

Comment: The age of this sample is consistent with early Middle Woodland period artifacts recovered from this context.

UTCAG07-003. McCrosky Is., Fea. 287, 40SV43C14-25 1100 ± 70
 $\delta^{13}\text{C} = -26.2\text{‰}$

Mixed charcoal sample recovered from a cylindrical pit feature. Collected 2006 and submitted 2007 by R R Polhemus. This feature is from the same context as features that provided samples 40SV43C14-17 and 40SV43C14-24. The age of this sample, however, is *not* consistent with early Middle Woodland period artifacts recovered from this context. Rather, this date indicates an early Mississippian period age for this feature.

Comment: Fine-scale analysis discovered the presence of small shell tempered ceramic sherds, indicating a Mississippian period intrusion into the feature.

CONCLUSIONS

The results obtained indicate that the sample pretreatment, benzene synthesis, counting vial standardization, and counting protocols adopted by the UTCAG ^{14}C laboratory are valid and produce comparably valid dating results. These conclusions are supported by the good agreement of the results of the 3 intercomparison samples and the good agreement between the dates and the identified archaeological contexts. The good agreement between the dates obtained for the 2 aliquots from 40SV43C14-18 also demonstrates that the vial standardization protocol is satisfactory. Overall, the results indicate that the UTCAG ^{14}C laboratory is operational and producing reliable results. This will be further tested by our participation in the Fifth International Radiocarbon Intercomparison (VIRI).

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