

COMETARY MOLECULES

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ABSTRACT. Which molecular species are firmly identified as cometary species and which are also interstellar? How may excitation effects bias the data interpretation?

1. Introduction

Because space is limited, only those molecules with observed and identified gas-phase spectra will be discussed. Thus, there can be no discussion of either the interesting mass spectra acquired by the Comet Halley spacecraft, or current work on the proposed polyoxymethylene identification (Huebner 1987). Modern theory links the composition of comets to that of the interstellar medium, with many uncertainties. Table 1 lists the known interstellar and circumstellar species by number of atoms, except for the polycyclic aromatic

TABLE 1. The 89 reported interstellar and circumstellar molecules as of July, 1991

2 AlF AlCl C₂ CH CH⁺ CN CO CP CS CSi HCl H₂
KCl NH NO NS NaCl OH PN SO SO⁺ SiN SiO SiS
3 C₂H C₂S HCN HCO HCO⁺ HCS⁺ H₃⁺ H₂O H₂S HNC
HNO N₂H⁺ OCS SO₂ c-SiC₂ 4 c-C₃H 1-C₃H C₃N
C₃O C₃S C₂H₂ HCNH⁺ H₂CO H₂CS H₃O⁺ HNCO HNCS
HOCO⁺ NH₃ 5 C₄H C₄Si c-C₃H₂ CH₂CN CH₄ HC₃N
HCOOH H₂C₂O H₂CHN H₂NCN SiH₄ 6 C₅H C₅O C₂H₄
CH₃CN CH₃NC CH₃OH CH₃SH HCONH₂ 7 C₆H CH₂CHCN
CH₃C₂H HC₅N HCOCH₃ NH₂CH₃ 8 CH₃C₃N HCOOCH₃
9 CH₃C₄H CH₃CH₂CN (CH₃)₂O CH₃CH₂OH HC₇N
10 CH₃C₅N (CH₃)₂CO 11 HC₉N 13 HC₁₁N

hydrocarbon (PAH) compounds. Ring structures are preceded by c-, but where the structure is also linear an l- is used. Mann and Williams (1980) and Lovas (1986; 1987) have prepared extensive frequency lists.

2. Cometary Molecules

The cometary molecules with observed and identified gas-phase spectra are listed in Table 2. The known interstellar/

TABLE 2. The 27 reported cometary molecules as of July 1991

<u>2</u>	C ₂	CH	CH ⁺	CN	CN ⁺	CO	CO ⁺	CS	N ₂ ⁺	NH	OH	OH ⁺	S ₂
<u>3</u>	HCN	H ₂ O	H ₂ O ⁺	H ₂ S	C ₃	CO ₂	CO ₂ ⁺	NH ₂	OCS				
<u>4</u>	H ₂ CO	NH ₃	<u>5</u>	CH ₄	<u>6</u>	CH ₃ CN	CH ₃ OH						

circumstellar molecules are more chemically complicated than these; along with the Halley mass spectra and meteoritic chemistry, this suggests that comets are more chemically evolved than their gas-phase spectra indicate.

Two important questions are: how secure is a given cometary molecular identification; and what molecules are identified in multiple spectral regions? Several reviews have addressed these questions (e.g., A'Hearn 1983; Feldman 1983; Wyckoff 1983; Snyder 1982), so this will be a brief update. The ultraviolet region will be defined as $\lambda < 3200 \text{ \AA}$, optical as $3200\text{--}8000 \text{ \AA}$, infrared as 8000 \AA to 0.5 mm (20 cm^{-1}), and radio as $\lambda > 0.5 \text{ mm}$ wavelength. Table 3 lists ultraviolet molecules, approximate band centers, and assignments. All are

TABLE 3. Cometary molecules with ultraviolet spectra

C ₂ (2313 Å, D ³ Σ _u ⁺ -X ¹ Σ _g ⁺)	CN ⁺ (2181 Å, f ¹ Σ-a ¹ Σ; 3185 Å, c ¹ Σ-a ¹ Σ)	CO (1510 Å, A ¹ Π-X ¹ Σ ⁺)	CO ⁺ (2190 Å, B ² Σ ⁺ -X ² Σ ⁺)
CS (2576 Å, A ¹ Π-X ¹ Σ ⁺)	OH (3090 Å, A ² Σ ⁺ -X ² Π _i)	S ₂ (2957 Å, B ³ Σ _u ⁻ -X ³ Σ _g ⁻)	CO ₂ ⁺ (2890 Å, B ² Σ _u ⁺ -X ² Π _g)

commonly found except for S₂, which was detected only in Comet IRAS-Araki-Alcock or I-A-A (A'Hearn 1983; Feldman 1983; 1991; Wyckoff 1983). Table 4 lists optical molecules, approximate band centers, and assignments (A'Hearn 1983; Wyckoff 1983).

TABLE 4. Cometary molecules with optical spectra

C ₂ (5165 Å, d ³ Π _g -a ³ Π _u ; 7715 Å, A ¹ Π _u -X ¹ Σ _g ⁺)	¹² C ¹³ C (4745 Å, d ³ Π _g -a ³ Π _u)	CH (3889 Å, B ² Σ ⁻ -X ² Π _r ;
4315 Å, A ² Δ-X ² Π _r)	CH ⁺ (4225 Å, A ¹ Π-X ¹ Σ ⁺)	
CN (3883 Å, B ² Σ ⁺ -X ² Σ ⁺ ; 7873 Å, A ² Π _i -X ² Σ ⁺)		
CO ⁺ (3954 Å, B ² Σ ⁺ -A ² Π _i ; 4273 Å, A ² Π _i -X ² Σ ⁺)		
N ₂ ⁺ (3914 Å, B ² Σ _u ⁺ -X ² Σ _g ⁺)	NH (3360 Å, A ³ Π _i -X ³ Σ ⁻)	
OH ⁺ (3565 Å, A ³ Π _i -X ³ Σ ⁻)	H ₂ O ⁺ (6198 Å, \tilde{A}^2A_1 - \tilde{X}^2B_1)	
C ₃ (4040 Å, $\tilde{A}^1\Pi_u$ - $\tilde{X}^1\Sigma_g^+$)	CO ₂ ⁺ (3509 Å, $\tilde{A}^2\Pi_u$ - $\tilde{X}^2\Pi_g$;	
3674 Å)	NH ₂ (5007 Å, \tilde{A}^2A_1 - \tilde{X}^2B_1)	

Table 5 lists cometary molecules observed in the infrared,

TABLE 5. Cometary molecules with infrared spectra

CO (4.7 $\mu\text{m}/2143 \text{ cm}^{-1}$, 1-0)
OH (1.5-1.9 $\mu\text{m}/6666-5263 \text{ cm}^{-1}$, 3-1, 4-2, 5-3, 12-9, 6-4, 7-5; 119.23 $\mu\text{m}/83.9 \text{ cm}^{-1}$, ${}^2\Pi_{3/2}$ $J=5/2-3/2, + -$)
H ₂ O (1.38 $\mu\text{m}/7249.8 \text{ cm}^{-1}$, $\nu_1+\nu_3$; 1.88 $\mu\text{m}/5331.3 \text{ cm}^{-1}$, $\nu_2+\nu_3$; 2.44 $\mu\text{m}/4100 \text{ cm}^{-1}$, $\nu_1+\nu_3-2\nu_2$ (tent. assignment); 2.64 $\mu\text{m}/3782.2 \text{ cm}^{-1}$, $\nu_2+\nu_3-\nu_2$; 2.66 $\mu\text{m}/3755.9 \text{ cm}^{-1}$, ν_3)
CO ₂ (4.3 $\mu\text{m}/2349 \text{ cm}^{-1}$, ν_3) OCS (4.85 $\mu\text{m}/2062 \text{ cm}^{-1}$, ν_3)
H ₂ CO (3.5 $\mu\text{m}/2843 \text{ cm}^{-1}$, ν_5 ; 3.6 $\mu\text{m}/2783 \text{ cm}^{-1}$, ν_1)
CH ₄ (3.3 $\mu\text{m}/3019 \text{ cm}^{-1}$, ν_3)

wavelengths (in $\mu\text{m}/\text{cm}^{-1}$), and assignments. Encrenaz and Knacke (1991) and Weaver et al. (1991) have given recent reviews. The 1.38 μm H₂O band was detected in Halley by Knacke et al. (1986) and Krasnopolsky et al. (1986), and the 1.88 μm band by Knacke et al. (1986). The 2.44 μm band is a tentative H₂O hot band assignment in Halley (Maillard et al. 1987), but the 2.64 μm assignment is definite (Weaver et al. 1986). The 2.66 μm assignment (ν_3) is well confirmed (Mumma et al. 1986; Weaver et al. 1986; Combes et al. 1988). Larson et al. (1989) also detected 10 H₂O emission lines from ν_3 and a hot band line from $\nu_2+\nu_3-\nu_2$ in Comet Wilson. CO, CO₂, OCS, and H₂CO were identified in Halley by Combes et al. (1988), but CO and OCS were weak. Stacey et al. (1987) detected OH in Halley at 119.23 μm and Krasnopolsky et al. (1986) detected the 1.5-1.9 μm OH bands. CH₄ in Halley was reported by Kawara et al. (1988) and in Wilson by Larson et al. (1989). Table 6 lists the molecules observed in the radio region, wavelengths/frequencies, and assignments. The radio CH line has been reported only in Comet Kohoutek (Black et al. 1974). Radio OH lines are routinely observed in comets. HCN detections in Kohoutek and Halley were followed by detections in Brorsen-Metcalf, Austin, and Levy (cf., Crovisier and Schloerb 1991; Colom et al. 1990; Schloerb and Ge 1990a;b). The 1.348 cm H₂O line was reported in comets Bradfield and I-A-A (Jackson et al. 1976; Altenhoff et al. 1983). H₂S and CH₃OH (3.099 and 2.066 mm) were detected in Austin (Bockelée-Morvan et al. 1991). CH₃OH (1.240 mm) also was detected in Levy (Schloerb and Ge 1990a). The 6.207 cm H₂CO line was detected in comets Halley and Machholz by Snyder et al. (1989; 1990), the 1.3283 mm line in Brorsen-Metcalf, Austin, and Levy by Colom et al. (1991), and the 0.8522 mm line in Levy by Schloerb and Ge (1990b). The 1.256 cm line of NH₃ was detected only in I-A-A (Altenhoff et al. 1983). Vibrationally excited CH₃CN (2.7079 mm) was observed only in Kohoutek (Ulich and Conklin 1974).

TABLE 6. Cometary molecules with radio spectra

 CH (8.988 cm/3,335.481 MHz, ${}^2\Pi_{1/2}$ J=1/2, F=1-1)
 OH (18.595 cm/1,612.2310 MHz ${}^2\Pi_{3/2}$ J=3/2, F=1-2;
 18.001 cm/1,665.4018 MHz ${}^2\Pi_{3/2}$ J=3/2, F=1-1;
 17.980 cm/1,667.3590 MHz ${}^2\Pi_{3/2}$ J=3/2, F=2-2;
 17.424 cm/1,720.5300 MHz ${}^2\Pi_{3/2}$ J=3/2, F=2-1)
 HCN (3.3825 mm/88,630.4157 MHz J=1-0, F=1-1;
 3.3824 mm/88,631.8473 MHz J=1-0, F=2-1; 3.3824 mm/
 88,633.9360 MHz J=1-0, F=0-1; 1.1285 mm/265,886.432
 MHz J=3-2; 0.8457 mm/354,505.472 MHz J=4-3)
 H₂O (1.348 cm/22,235.120 MHz $J_{KaKc}=6_{16}-5_{23}$, F=5-4)
 H₂S (1.7764 mm/168,762.762 MHz $J_{KaKc}=1_{10}-1_{01}$)
 H₂CO (6.207 cm/4829.6639 MHz $J_{KaKc}=1_{11}-1_{10}$, F=2-1;
 1.3283 mm/225,697.775 MHz $J_{KaKc}=3_{12}-2_{11}$;
 0.8522 mm/351,768.639 MHz $J_{KaKc}=5_{15}-4_{14}$)
 NH₃ (1.256 cm/23870.1296 MHz J(K)=3(3), F=4-4)
 CH₃CN (2.7079 mm/110,709.55 MHz $\nu_8=1$, J(K)=6(3)-5(3);
 2.7079 mm/110,712.22 MHz $\nu_8=1$, J(K)=6(0)-5(0))
 CH₃OH (3.0990 mm/96,739.39 MHz J(K)=2(-1)-1(-1)E;
 3.0989 mm/96,741.42 MHz J(K)=2(0)-1(0)A+; 2.0662 mm/
 145,093.75 MHz J(K)=3(0)-2(0)E; 2.0661 mm/145,097.47
 MHz J(K)=3(-1)-2(-1)E; 2.0661 mm/145,103.23 MHz
 J(K)=3(0)-2(0)A+; 2.0657 mm/145,131.88 MHz J(K)=
 3(1)-2(1)E; 1.2400 mm/241,767.224 MHz J(K)=5(-1)-
 4(-1)E; 1.2399 mm/241,791.431 MHz J(K)=5(0)-4(0)A+)

3. Excitation Effects in Comets

In the ultraviolet and optical spectral regions, the dominant excitation mechanism for electronic transitions is resonance fluorescence. The Doppler-shifted resonant transitions of cometary molecules absorb the solar spectrum and reradiate it via various electronic transitions. Typically, collisions play only a minor role in establishing the molecular population distributions in these spectra (cf., Arpigny 1976). In the longer wavelength spectral regions, a major excitation mechanism is believed to be infrared fluorescence, but radio excitation may be more complicated.

Because ultraviolet excitation mechanisms are fairly well understood, the detection of S₂ in I-A-A, as discussed by M. A'Hearn (this book), can be attributed both to the existence of an outburst at closest approach and to the remarkable 25 km/arcsec scale of the comet.

The intensities of the radio OH lines are determined by ultraviolet pumping called the Swings effect (cf., Schloerb and Gerard 1985; Crovisier and Schloerb 1991; de Pater et al. 1991). Briefly, solar ultraviolet Fraunhofer radiation pumps the cometary OH ultraviolet bands listed in Table 3 (3090 Å, A²Σ⁺-X²Π₁). The OH cascades to the ground state, establishing

the intensities of the 18 cm lines in Table 6. This effect determines whether radio OH will be observed in maser emission, absorption, or not at all! For example, when the projected densities of the unsplit upper and lower levels of the ground-state A doublet are almost equal, the cometary OH signal may vanish, even though the OH has not. Another point is that single radio antennas can make crude maps of cometary OH with a resolution of a few arcminutes, but radio interferometers can produce detailed imaging which can greatly affect the excitation modeling (cf., de Pater et al. 1991).

The evidence for an extended cometary dust coma that began to emerge with the measurements of Comet Halley is:

A. The in situ measurements made with the Giotto neutral gas mass spectrometer by Eberhardt et al. (1987) can be explained if CO is produced both directly from the nucleus and from extended dust in the inner coma.

B. The $>60,000$ km CN and C_2 jets observed by A'Hearn et al. (1986a,b) can be explained by dissociation directly from CHON particles in an extended dust coma.

C. The VLA detected the 6.207 cm line of H_2CO in emission from Halley and Machholz (Snyder et al. 1989; 1990). Interpreting this line requires an extended coma model.

D. The cometary H_2CO model of Bockelée-Morvan and Crovisier (1991) shows how H_2CO will behave if produced only by the cometary nucleus, with no further gas production in an extended gas/dust coma. This is an excellent model, but it does not explain (a) the VLA detections of the 6.207 cm H_2CO line; (b) the IRAM detections of the 1.3283 mm H_2CO line at a nuclear distance of 3500 km from Austin and 6300 km from Levy by Colom et al. (1991); or (c) the CSO detection of the 0.8522 mm H_2CO line at a 30" nuclear distance offset from Comet Levy by Schloerb (1991, private communication). These observations suggest that either specious comet lines are extremely common at different frequencies and times, or there is indeed an extended cometary dust coma which needs to be taken into account in future models.

4. Summary

Newly detected species include H_2O , H_2CO , CH_4 (Halley), CH_3OH (Austin and Levy), and H_2S (Austin). Cometary nuclei appear to be more chemically evolved than the current gas-phase spectroscopy suggests. Some of the excitation conditions are exemplified by the sudden appearance of S_2 in I-A-A, the ultraviolet pumping of radio OH lines, the extended dust coma, and the unexplained radio lines of H_2O and excited CH_3CN .

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QUESTIONS AND ANSWERS

J.P.Maillard: You listed CO as a detected molecule in the IR (at $4.7 \mu m$). As far I know it has not been detected. It has been looked for, of course, but only upper limit has been obtained.

L.E.Snyder: CO was identified in Halley by Combes et al (1988), but it was weak.

T.J.Millar: If cometary particles are related to interstellar particles, one might expect them to contain significant fractionation in deuterium. Have there been sensitive searches for D-bearing molecules in comets?

L.E.Snyder: This question was answered by Mike A'Hearn.

T.J.Millar: What is the abundance of D?

M.F.A'Hearn: The Giotto NMS was used in its ion mode by Eberhardt to derive separate upper & lower limits on HDO/ H_2O in Halley at something like 5×10^{-5} to 5×10^{-4} (I don't remember the exact numbers). Spectra from IUE have been searched for emission by OD and upper limits have been derived for 5 - 10 comets ranging from 10^{-2} to 4×10^{-4} , depending on signal-to-noise.